



*The Chinese University of Hong Kong  
Department of Chemistry  
Research Seminar Series*

**Speaker:** Professor George A. O'Doherty  
Department of Chemistry  
Northeastern University

**Title:** De Novo Asymmetric Synthesis of Natural  
Products for Stereochemical Structure  
Activity Relationship (S-SAR) Studies

**Date:** May 9, 2018 (Wednesday)

**Time:** 2:30 p.m.

**Venue:** L3  
Science Centre





*The Chinese University of Hong Kong*  
*Department of Chemistry*  
*Research Seminar Series*

**Speaker:** Professor Tiow-Gan Ong  
Institute of Chemistry, Academia Sinica, Taipei

**Title:** Carbene and Catalysis

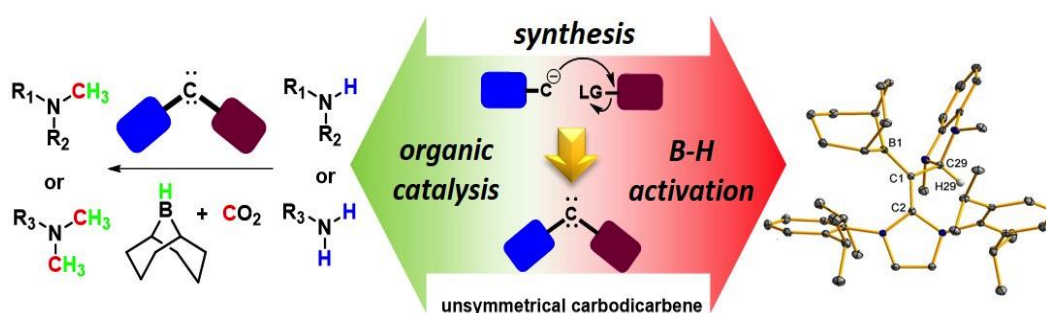
**Date:** May 10, 2018 (Thursday)

**Time:** 10:30 a.m.

**Venue:** L3, Science Centre

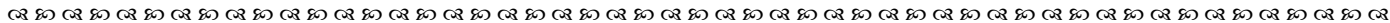
< Abstract >

Carbene is a special class of carbon compound with a structural  $CL_2$  that feature a dicoordinated central carbon (0) bearing two lone pairs of electrons flanked by dative ligands (L) units. When L is an NHC or carbene, a particular carbene with a carbon→carbon dative bond is formed. This carbene species is classified as a carbodicarbene (CDC) with a formulation of  $NHC \rightarrow C \leftarrow NHC$ . Due to the two lone pairs on the central carbon, carbene or carbodicarbene has a strong  $\sigma$ -donating scaffold. In describe the development of a synthetic protocol for the preparation of carbodicarbenes with unsymmetrical side arms and investigate their late transition metal complexes and intrinsic reactivities. The ligand carbodicarbene is used for possible catalysis application and main group chemistry.



**Reference**

1. W.-C. Chen, W.-C. Shih, T. Jurca, L. Zhao, D. M. Andrada, C.-J. Peng, C.-C. Chang, S.-k. Liu, Y.-P. Wang, Y.-S. Wen, G. P. A. Yap, C.-P. Hsu, G. Frenking\*, **T.-G. Ong**\* *J. Am. Chem. Soc.* **2017**, *139*, 12830-12836.
2. Chen, W.-C.; Shen, J.-S.; Jurca, T.; Peng, C.-J.; Lin, Y. H.; Wang, Y.-P.; Shih, W.-C.; Yap, G. P. A.; **Ong, T.-G.**\* *Angew. Chem. Int. Ed.*, **2015**, *54*, 15307-15212.
3. Hsu, Y.-C.; Shen, J.-S.; Lin, B.-C.; Chen, W.-C.; Chan, Y.-T.; Ching, W.-M.; Yap, G. P. A.; Hsu, C.-P.; **Ong, T.-G.**\* *Angew. Chem. Int. Ed.*, **2015**, *54*, 2420-2424.
4. Chen, C.-C.; Lee, C.-Y.; Lin, B.-C.; Hsu, Y.-C.; Shen, J.-S.; Hsu, C.-P.; Yap, G. P. A.; **Ong, T.-G.**\* *J. Am. Chem. Soc.*, **2014**, *136*, 914-917.



*The Chinese University of Hong Kong*  
*Department of Chemistry*  
*Research Seminar Series*

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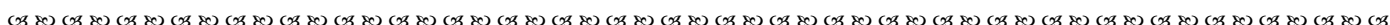
**Speaker:** Professor Jia Xie  
School of Electrical and Electronic Engineering  
Huazhong University of Science and  
Technology

**Title:** Natural polymer based energy storage  
materials and devices

**Date:** May 11, 2018 (Friday)

**Time:** 4:30 p.m.

**Venue:** L1  
Science Centre



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Contact Person:  
Prof. Jiang Xia

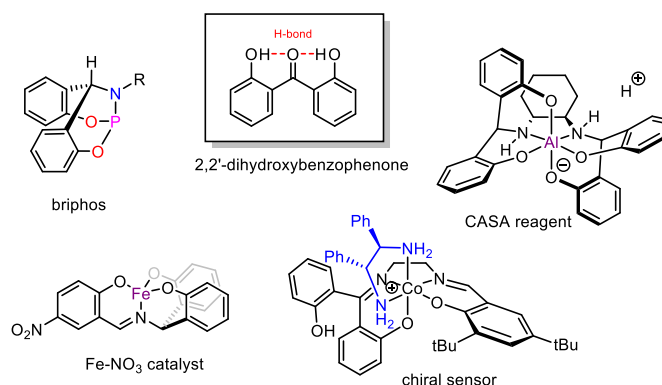


**The Chinese University of Hong Kong**  
**Department of Chemistry**  
**Research Seminar Series**

- Speaker:** Professor Hyunwoo Kim  
 Department of Chemistry  
 Korea Advanced Institute of Science and Technology (KAIST)
- Title:** Geometrical Constraints in Design of Metal Ligands and Chiral Sensors
- Date:** May 18, 2018 (Friday)
- Time:** 4:30 p.m.
- Venue:** L1, Science Centre

< Abstract >

The so-called privileged ligands have been creatively used to enhance the catalytic performance of various transition metals. The ultimate goal of my research is to develop another class of privileged ligands for organo- and transition-metal catalysts. During the last several years of research, we have observed a remarkable reactivity of 2,2'-dihydroxybenzophenone, which is based on a new phosphorus ligand (briphos) and chiral sensors (CASA reagents). In order to explain the unexpected electronic properties of briphos ligands, we have proposed a new concept, geometric control, to modulate the ligand property in addition to conventional steric and electronic control. We have shown that the briphos ligand can be a tunable and scalable  $\pi$ -acceptor ligands for Rh-catalyzed conjugate additions of boronic acids and Pd-catalyzed dehydrative couplings of allylic alcohols. We also have reported that negatively charged octahedral Al complexes can be conveniently used for  $^1\text{H}$  NMR chiral analysis of various charged molecules including amines and carboxylic acids as well as alcohols. In addition, a new Fe- $\text{NO}_3$  catalyst was developed and used for selective formation of cyclic carbonates from internal epoxides and  $\text{CO}_2$ .



#### References

1. Seo, M.-S.; Jang, S.; Kim, H.\* *Chem. Commun.*, **2018**, ASAP.
2. Sinha, I.; Lee, Y.; Bae, C.; Tussupbayev, S.; Lee, Y.; Seo, M.-S.; Kim, J.; Baik, M.-H.\*; Lee, Y.\*; Kim, H.\* *Catal. Sci. Technol.* **2017**, 7, 4375.
3. Jung, H.; Lee, A.; Kim, J.; Kim, H.\*; Baik, M.-H.\* *Adv. Synth. Catal.* **2017**, 359, 3160.
4. Lee, A.; Kim, H.\* *J. Org. Chem.* **2016**, 81, 3520.
5. Kang, J.; Kim, J.; Lee, A.; Kim, W. Y.; Kim, H.\* *Org. Lett.* **2016**, 18, 616.
6. Seo, M. S.; Kim, H.\* *J. Am. Chem. Soc.* **2015**, 137, 14190.
7. Lee, A.; Kim, H.\* *J. Am. Chem. Soc.* **2015**, 137, 11250.
8. Lee, A.; Ahn, S.; Kang, K.; Seo, M.-S.; Kim, Y.; Kim, W. Y.\*; Kim, H.\* *Org. Lett.* **2014**, 16, 5490.



*The Chinese University of Hong Kong*  
**Seminar**

Jointly Organized by  
Department of Chemistry  
School of Life Science  
(普通話主講)

**Speaker:** Professor Jianji Wang (王鍵吉教授)  
School of Chemistry and Chemical  
Engineering (化學化工學院)  
Henan Normal University (河南師範大學)

**Title:** CO<sub>2</sub> 响應離子液體的設計及應用

**Date:** 21 May, 2018 (Monday)

**Time:** 2:30 p.m.

**Venue:** L3  
Science Centre



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Contact Person:  
Prof. Zuowei Xie



*The Chinese University of Hong Kong*  
*Department of Chemistry*  
*Research Seminar Series*

**Speaker:** Professor Kazuaki Ishihara  
 Graduate School of Engineering  
 Nagoya University

**Title:** Rational Design of High Performance Catalysts Based on  
 Acid–Base Combination Chemistry

<< Abstract >>

We have studied on rational design of high performance catalysts based on acid–base combination chemistry. In this lecture, two topics are focused. One is “Cooperative catalytic system of chiral Lewis base catalysts and Lewis acids for enantioselective halocyclizations.”<sup>1</sup> The other is “Highly active chiral strong Brønsted acid–base salt catalysts for several enantioselective reactions.”<sup>2</sup>

**References**

1. (a) Sakakura, A.; Ukai, A.; Ishihara, K. *Nature* **2007**, *445*, 900. (b) Sawamura, Y.; Nakatsuji, H.; Sakakura, A.; Ishihara, K. *Chem. Sci.* **2013**, *4*, 4181. (c) Nakatsuji, H.; Sawamura, Y.; Sakakura, A.; Ishihara, K. *Angew. Chem. Int. Ed.* **2014**, *53*, 6974. (d) Lu, Y.; Nakatsuji, H.; Okumura, M.; Yao, L.; Ishihara, K. to be submitted.
2. (a) Hatano, M.; Maki, T.; Moriyama, K.; Arinobe, M.; Ishihara, K. *J. Am. Chem. Soc.* **2008**, *130*, 16858. (b) Hatano, M.; Hattori, Y.; Furuya, Y.; Ishihara, K. *Org. Lett.* **2009**, *11*, 2321. (c) Hatano, M.; Sugiura, Y.; Akakura, M.; Ishihara, K. *Synlett* **2011**, 1247. (d) Hatano, M.; Ozaki, T.; Sugiura, Y.; Ishihara, K. *Chem. Commun.* **2012**, *48*, 4986. (e) Hatano, M.; Ozaki, T.; Nishikawa, K.; Ishihara, K. *J. Org. Chem.* **2013**, *78*, 10405. (f) Hatano, M.; Ishihara, K. *Asian J. Org. Chem.* **2014**, *3*, 352. (g) Hatano, M.; Nishikawa, K.; Ishihara, K. *J. Am. Chem. Soc.* **2017**, *139*, 8424. (h) Hatano, M.; Mochizuki, T.; Nishikawa, K.; Ishihara, K. *ACS Catal.* **2018**, *8*, 349.

**Date:** May 28, 2018 (Monday)

**Time:** 10:30 a.m.

**Venue:** L3, Science Centre



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Contact Person:  
 Prof. Y.Y. Yeung



*The Chinese University of Hong Kong*  
*Department of Chemistry*  
*Research Seminar Series*

**Speaker:** Professor Kazuaki Ishihara  
 Graduate School of Engineering  
 Nagoya University

**Title:** Rational Design of High Performance Catalysts Based on  
 Acid–Base Combination Chemistry

<< Abstract >>

We have studied on rational design of high performance catalysts based on acid–base combination chemistry. In this lecture, two topics are focused. One is “Cooperative catalytic system of chiral Lewis base catalysts and Lewis acids for enantioselective halocyclizations.”<sup>1</sup> The other is “Highly active chiral strong Brønsted acid–base salt catalysts for several enantioselective reactions.”<sup>2</sup>

**References**

1. (a) Sakakura, A.; Ukai, A.; Ishihara, K. *Nature* **2007**, *445*, 900. (b) Sawamura, Y.; Nakatsuji, H.; Sakakura, A.; Ishihara, K. *Chem. Sci.* **2013**, *4*, 4181. (c) Nakatsuji, H.; Sawamura, Y.; Sakakura, A.; Ishihara, K. *Angew. Chem. Int. Ed.* **2014**, *53*, 6974. (d) Lu, Y.; Nakatsuji, H.; Okumura, M.; Yao, L.; Ishihara, K. to be submitted.
2. (a) Hatano, M.; Maki, T.; Moriyama, K.; Arinobe, M.; Ishihara, K. *J. Am. Chem. Soc.* **2008**, *130*, 16858. (b) Hatano, M.; Hattori, Y.; Furuya, Y.; Ishihara, K. *Org. Lett.* **2009**, *11*, 2321. (c) Hatano, M.; Sugiura, Y.; Akakura, M.; Ishihara, K. *Synlett* **2011**, 1247. (d) Hatano, M.; Ozaki, T.; Sugiura, Y.; Ishihara, K. *Chem. Commun.* **2012**, *48*, 4986. (e) Hatano, M.; Ozaki, T.; Nishikawa, K.; Ishihara, K. *J. Org. Chem.* **2013**, *78*, 10405. (f) Hatano, M.; Ishihara, K. *Asian J. Org. Chem.* **2014**, *3*, 352. (g) Hatano, M.; Nishikawa, K.; Ishihara, K. *J. Am. Chem. Soc.* **2017**, *139*, 8424. (h) Hatano, M.; Mochizuki, T.; Nishikawa, K.; Ishihara, K. *ACS Catal.* **2018**, *8*, 349.

**Date:** May 28, 2018 (Monday)

**Time:** 10:30 a.m.

**Venue:** L3, Science Centre



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Contact Person:  
 Prof. Y.Y. Yeung



*The Chinese University of Hong Kong*  
*Department of Chemistry*  
*Research Seminar Series*

**Speaker:** Professor Kazuaki Ishihara  
 Graduate School of Engineering  
 Nagoya University

**Title:** Rational Design of Amidation and Esterification Catalysts Based on Acid–Base Combination Chemistry

<< Abstract >>

We have studied on rational design of high performance catalysts based on acid–base combination chemistry. In this general lecture, two topics are focused. One is “Boronic Acid Catalysts for Dehydrative Condensation between Carboxylic Acids and Amines.”<sup>1,2</sup> The other is “Acid–base Combined Catalysts for Esterification (acylation, transesterification, and dehydrative condensation).”<sup>1,3–6</sup>

**References**

1. For a review article, see: Ishihara, K. *Tetrahedron* **2009**, *65*, 1085 (*Tetrahedron Report*).
2. For Boronic acid catalysts, see: (a) Ishihara, K.; Ohara, S.; Yamamoto, H. *J. Org. Chem.* **1996**, *61*, 4196. (b) Sakakura, A.; Ohkubo, T.; Yamashita, R.; Akakura, M.; Ishihara, K. *Org. Lett.* **2011**, *13*, 892. (c) Yamashita, R.; Sakakura, A.; Ishihara, K. *Org. Lett.* **2013**, *15*, 3654. (d) Lu, Y.; Wang, K.; Ishihara, K. *Asian J. Org. Chem.* **2017**, *6*, 1111. (e) Ishihara, K.; Lu, Y. *Chem. Sci.* **2017**, *7*, 1276. (f) Wang, K.; Lu, Y., Ishihara, K. *Chem Commun.* **2018**, just accepted (DOI: 10.1039/C8CC02558D).
3. For Hf (IV) and Zr(IV) catalysts for dehydrative condensation, see: (a) Ishihara, K.; Ohara, S.; Yamamoto, H. *Science* **2000**, *290*, 1140. (b) Ishihara, K.; Nakayama, M.; Ohara, S.; Yamamoto, H. *Tetrahedron* **2002**, *58*, 8179.
4. For Brønsted acid catalysts for dehydrative condensation, see: (a) Ishihara, K.; Nakagawa, S.; Sakakura, A. *J. Am. Chem. Soc.* **2005**, *127*, 4168. (b) Sakakura, A.; Koshikari, Y.; Ishihara, K. *Tetrahedron Lett.* **2008**, *49*, 5017. (c) Sakakura, A.; Koshikari, Y.; Akakura, M.; Ishihara, K. *Org. Lett.* **2012**, *14*, 30. (d) Koshikari, Y.; Sakakura, A.; Ishihara, K. *Org. Lett.* **2012**, *14*, 3194.
5. For acylation catalysts and asymmetric acylation catalysts, see: (a) Sakakura, A.; Kawajiri, K.; Ohkubo, T.; Kosugi, Y.; Ishihara, K. *J. Am. Chem. Soc.* **2007**, *129*, 14775. (b) Ishihara, K.; Kosugi, Y.; Akakura, M. *J. Am. Chem. Soc.* **2004**, *126*, 12212. (c) Ishihara, K.; Kosugi, Y.; Umemura, S.; Sakakura, A. *Org. Lett.* **2008**, *10*, 3191.
6. For transesterification catalysts, see: (a) Ishihara, K.; Niwa, M.; Kosugi, Y. *Org. Lett.* **2008**, *10*, 2187. (b) Hatano, M.; Furuya, Y.; Shimmura, T.; Moriyama, K.; Kamiya, S.; Maki, T.; Ishihara, K. *Org. Lett.* **2011**, *13*, 426. (c) Hatano, M.; Kamiya, S.; Moriyama, K.; Ishihara, K. *Org. Lett.* **2011**, *13*, 430. (d) Hatano, M.; Kamiya, S.; Ishihara, K. *Chem. Commun.* **2012**, *48*, 9465. (e) Hatano, M.; Ishihara, K. *Chem. Commun.* **2013**, *49*, 1983. (f) Hatano, M.; Tabata, Y.; Yoshida, Y.; Toh, K.; Yamashita, K.; Ogura, Y.; Ishihara, K. *Green Chem.* **2018**, *20*, 1193.

**Date:** May 29, 2018 (Tuesday)

**Time:** 2:30 p.m.

**Venue:** L1, Science Centre



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Contact Person:  
 Prof. Y.Y. Yeung