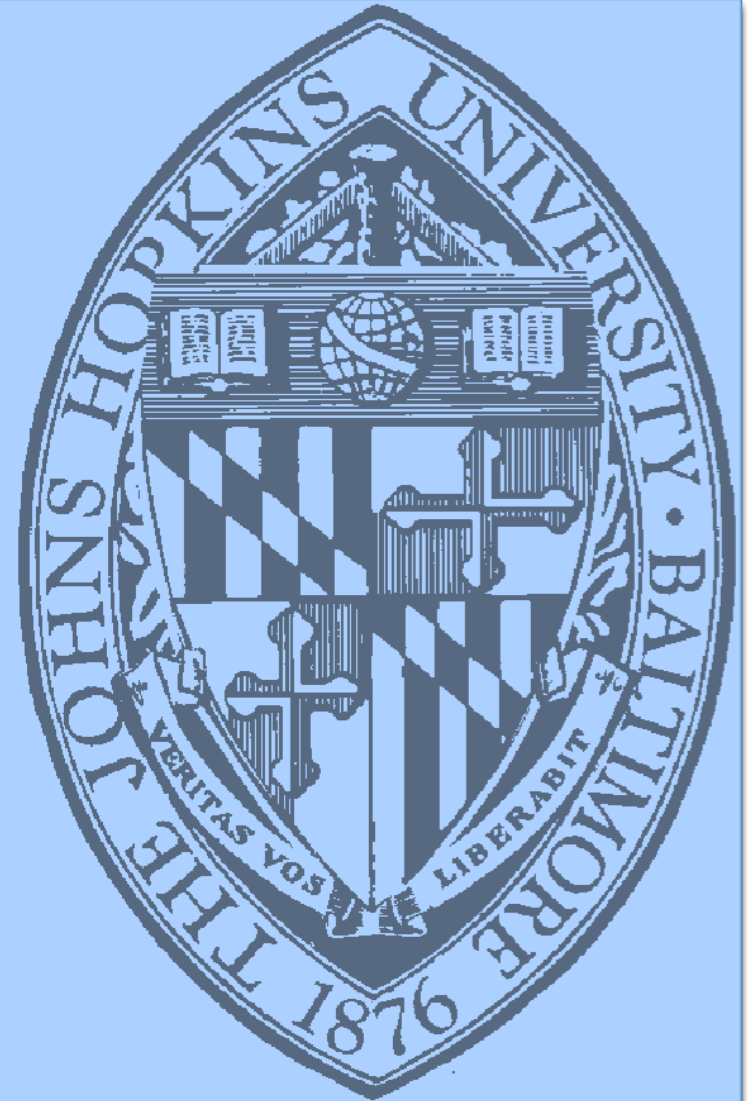


Plastic Deformation in Amorphous Materials

Michael L. Falk

***Materials Science and Engineering
Whiting School of Engineering
Johns Hopkins University***



My Thread This Week

Mon
Intro to
Mechanics
of Materials

Tues
Intro to
Molecular
Dynamics

Wed
Plasticity in
Amorphous
Materials

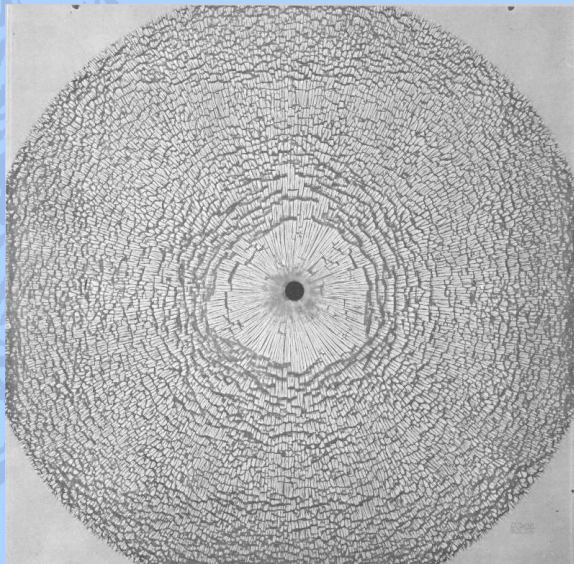
Thurs
STZ
Constitutive
Theory

Amorphous Materials (Glass)

- Represent a broad class of useful metal, ceramic and polymer materials in current application and in development. Their common characteristic is a lack of crystalline order.



PMMA house wares



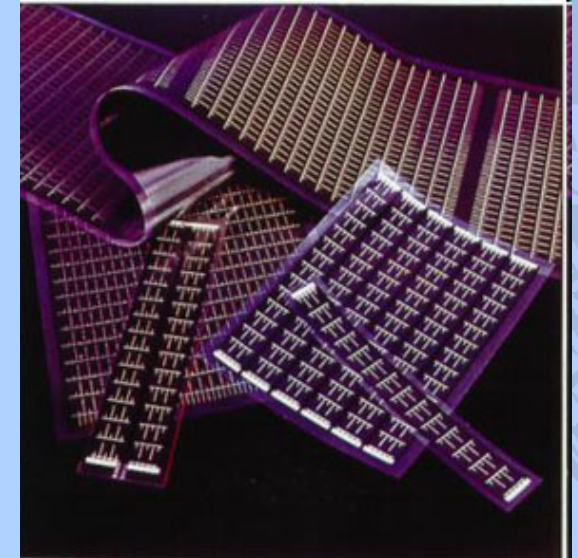
Safety Glass Cracking,
Harold Edgerton, 1938

10 Sept 08



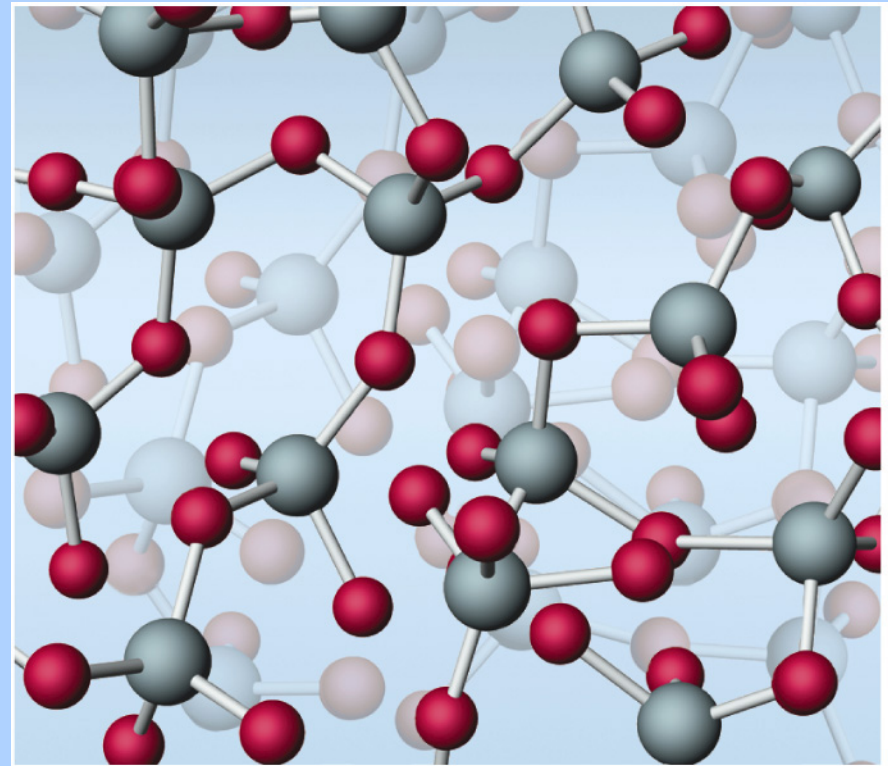
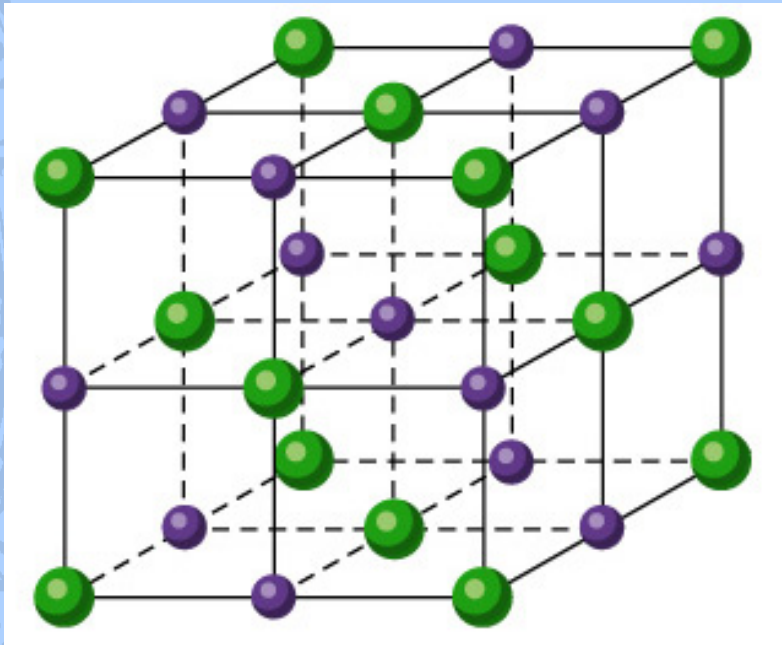
Metallic Glass Golf Club
Head, Liquidmetal Tech.

CUHK-ITP Mini Workshop



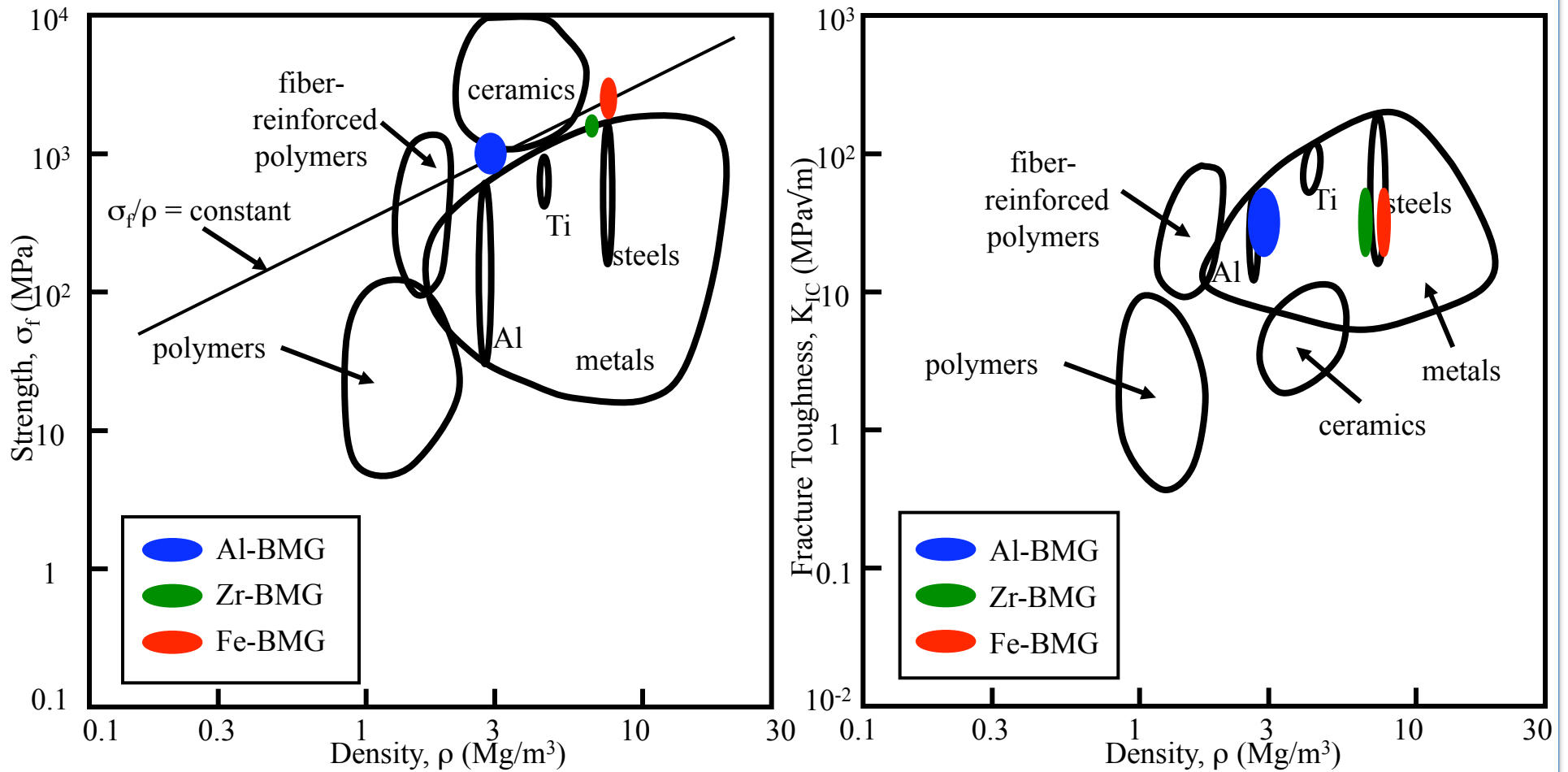
Flexible Si Solar Cells,
Iowa Thin Film Tech.

Orderly Crystals vs. Disorderly Glass

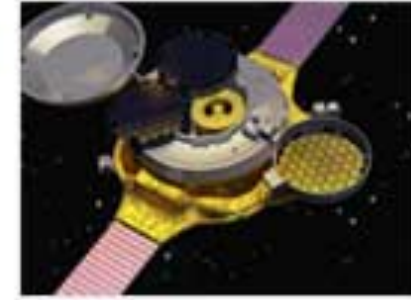


Metallic Glass

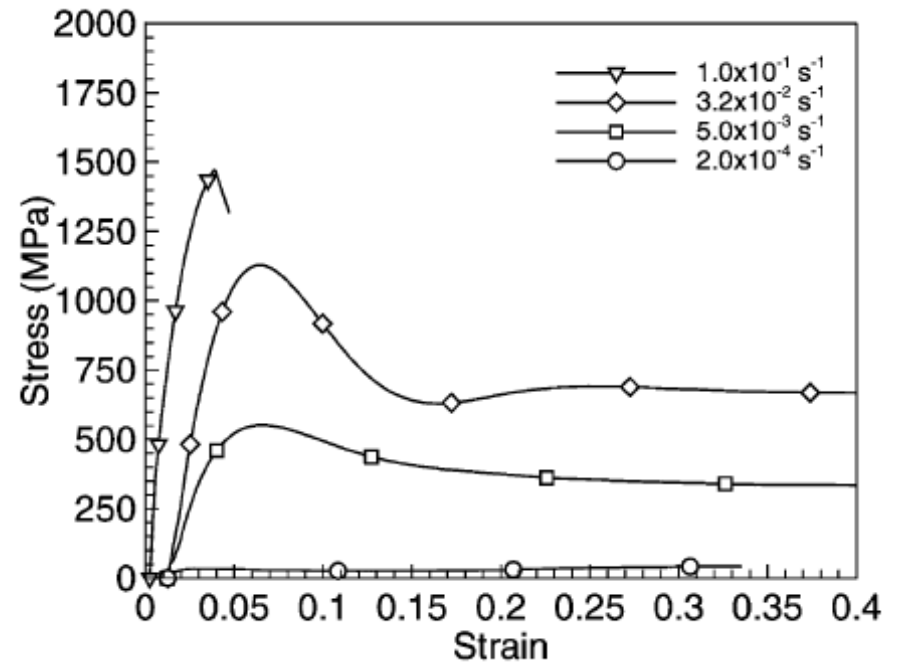
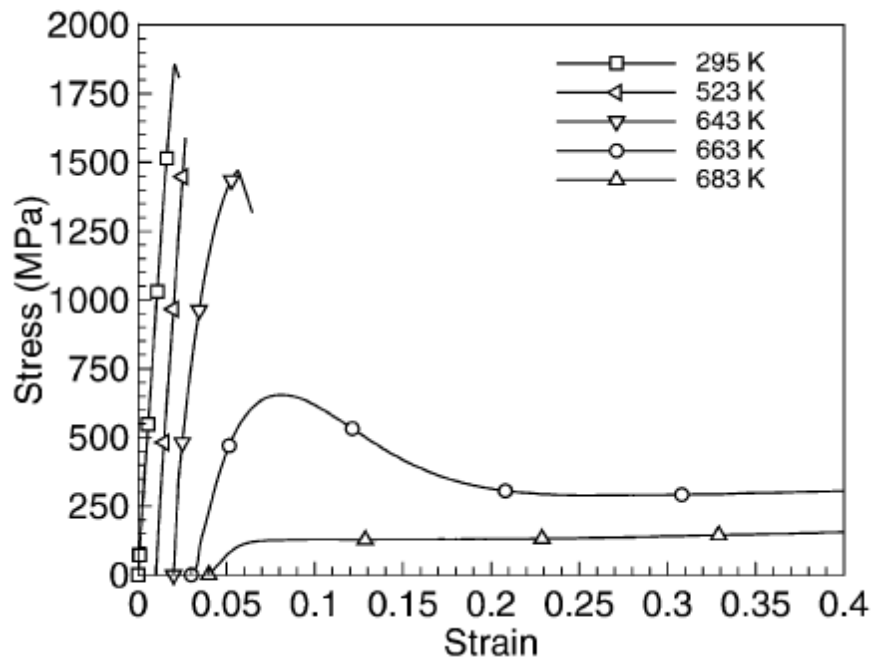
graphs courtesy of Katherine Flores, OSU



Applications of Bulk Metallic Glasses

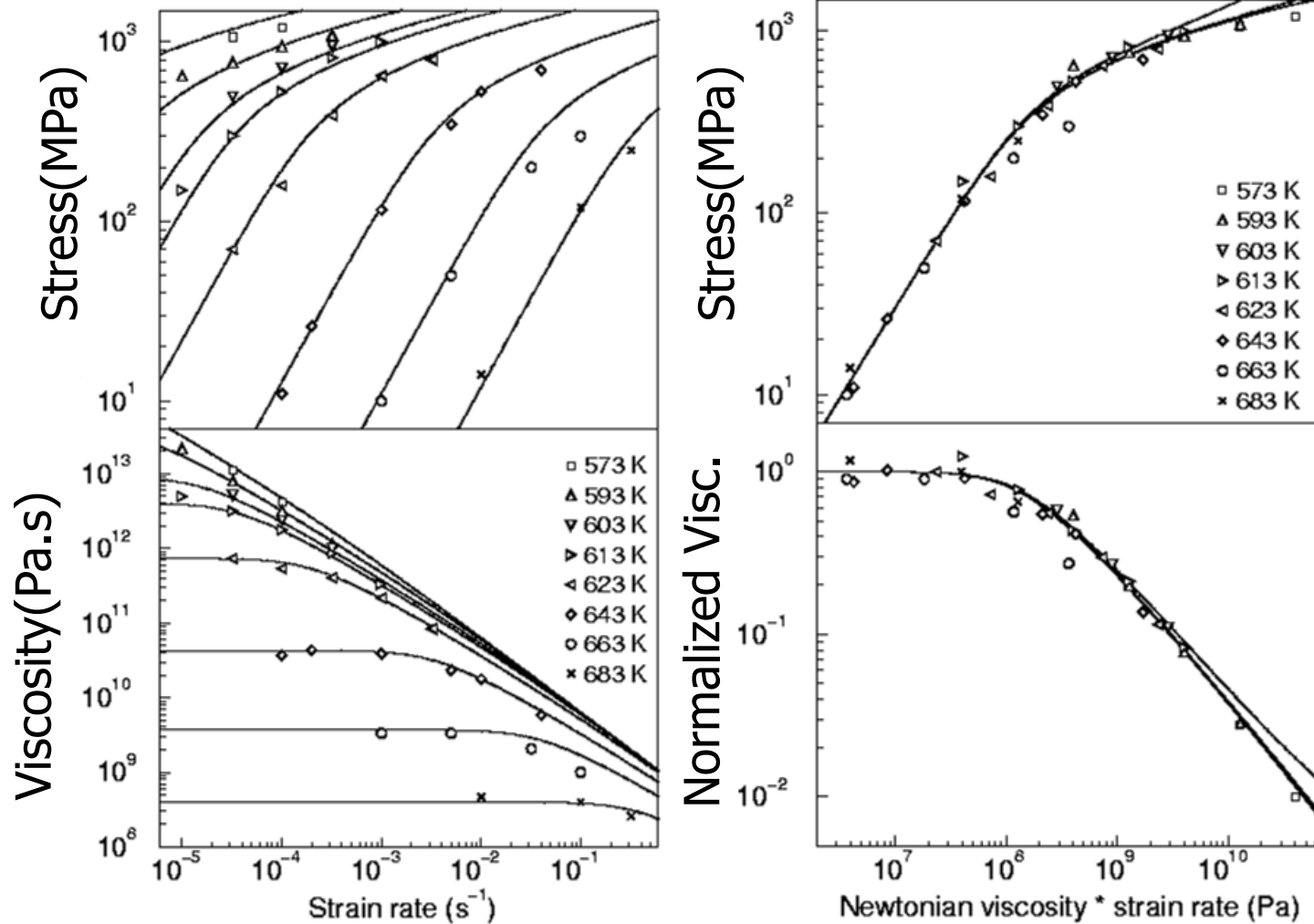


Mechanical Response Near T_g



Lu, Ravichandran, Johnson Acta Mat. 51, 3429 (2003)

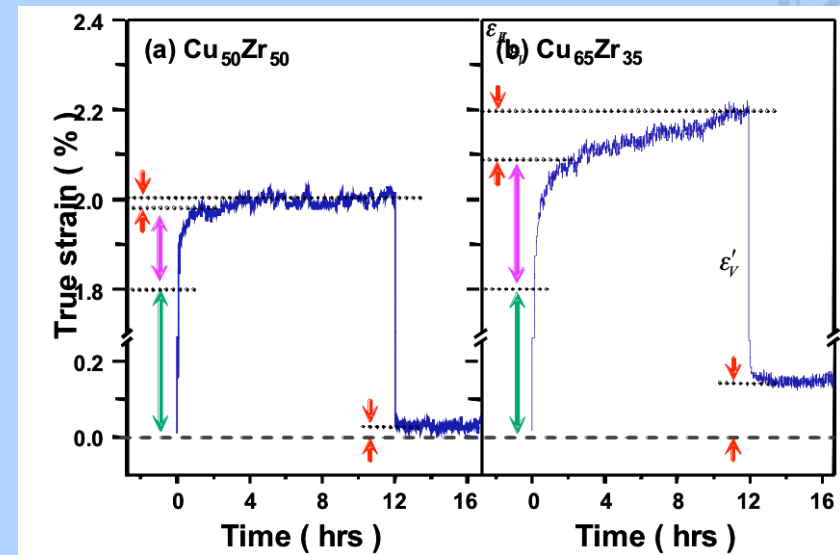
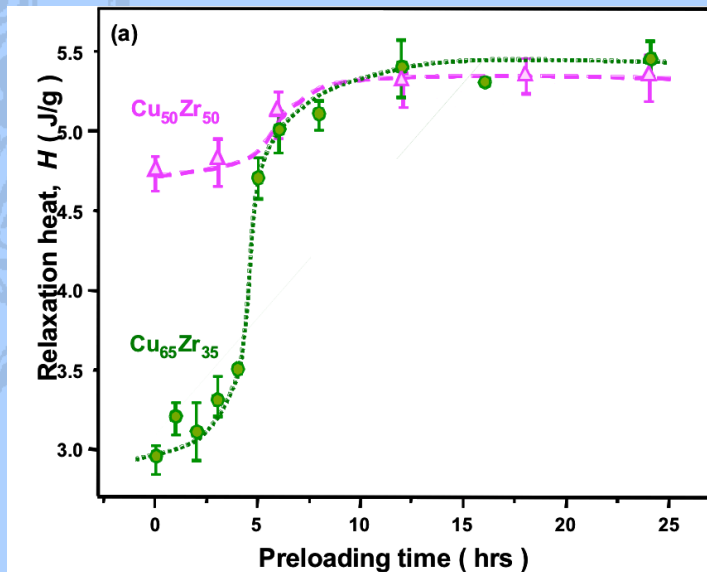
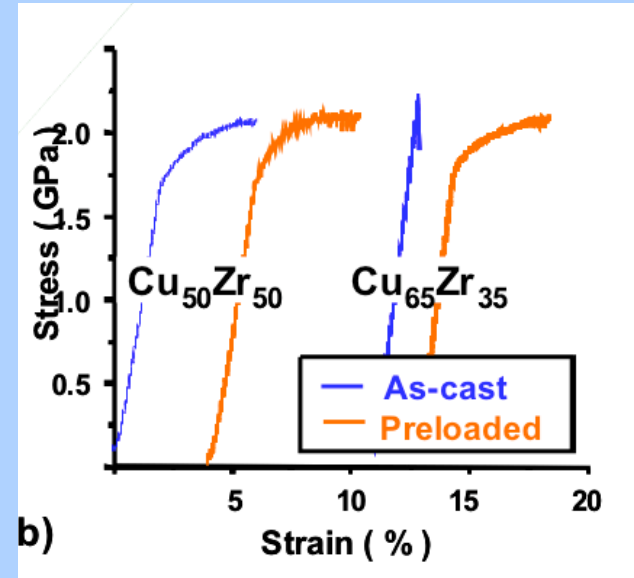
Rheology Near T_g



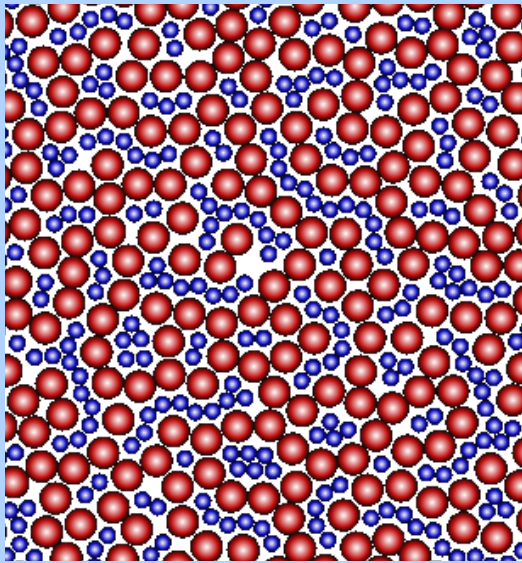
Lu, Ravichandran, Johnson Acta Mat. 51, 3429 (2003)

Behavior at Room Temperature

- Recent work in collaboration with Jae-Chul Lee at Korea University shows apparent contradiction
- Alloys that exhibit a large degree of plasticity when loaded at constant stress, show low ductility loaded at constant strain rate
- This appears to be related to structural changes during shear

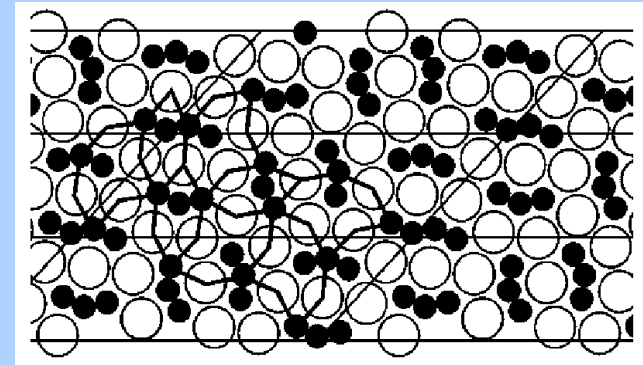


2D Simulation System

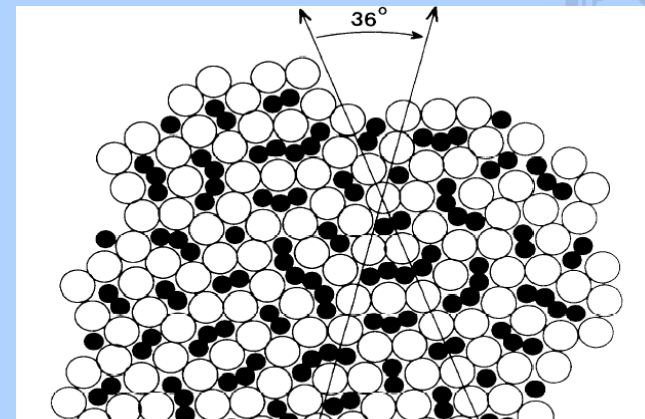


(Lancon et al, Europhys. Lett, 1986)

- 2D binary Lennard-Jones 12-6 potential
 - Binary system with quasi-crystalline packing
- 45:55 composition, 20,000-80,000 atoms
- $T_{MCT} \approx 0.325$

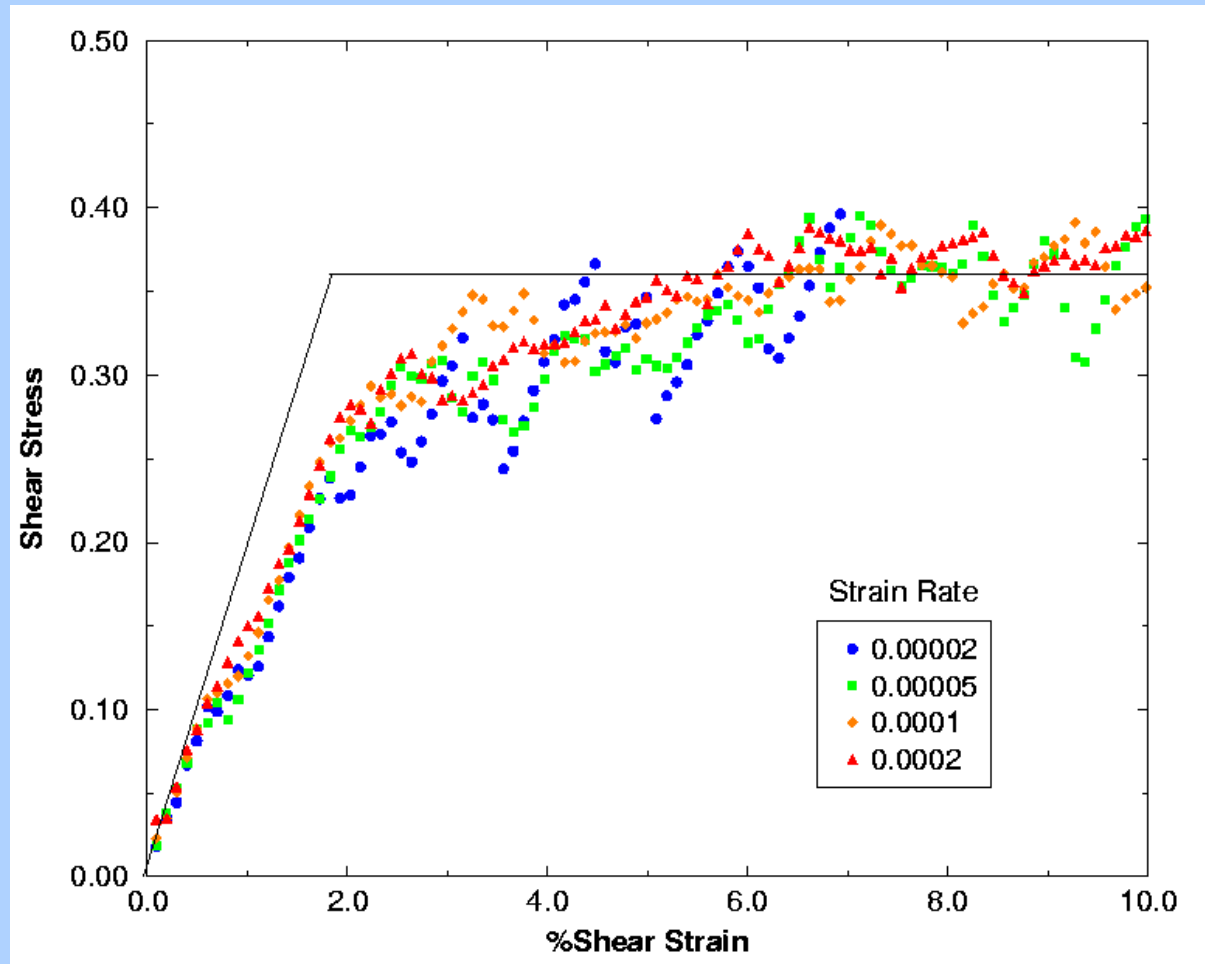


Lee, Swendsen, Widom (2001)



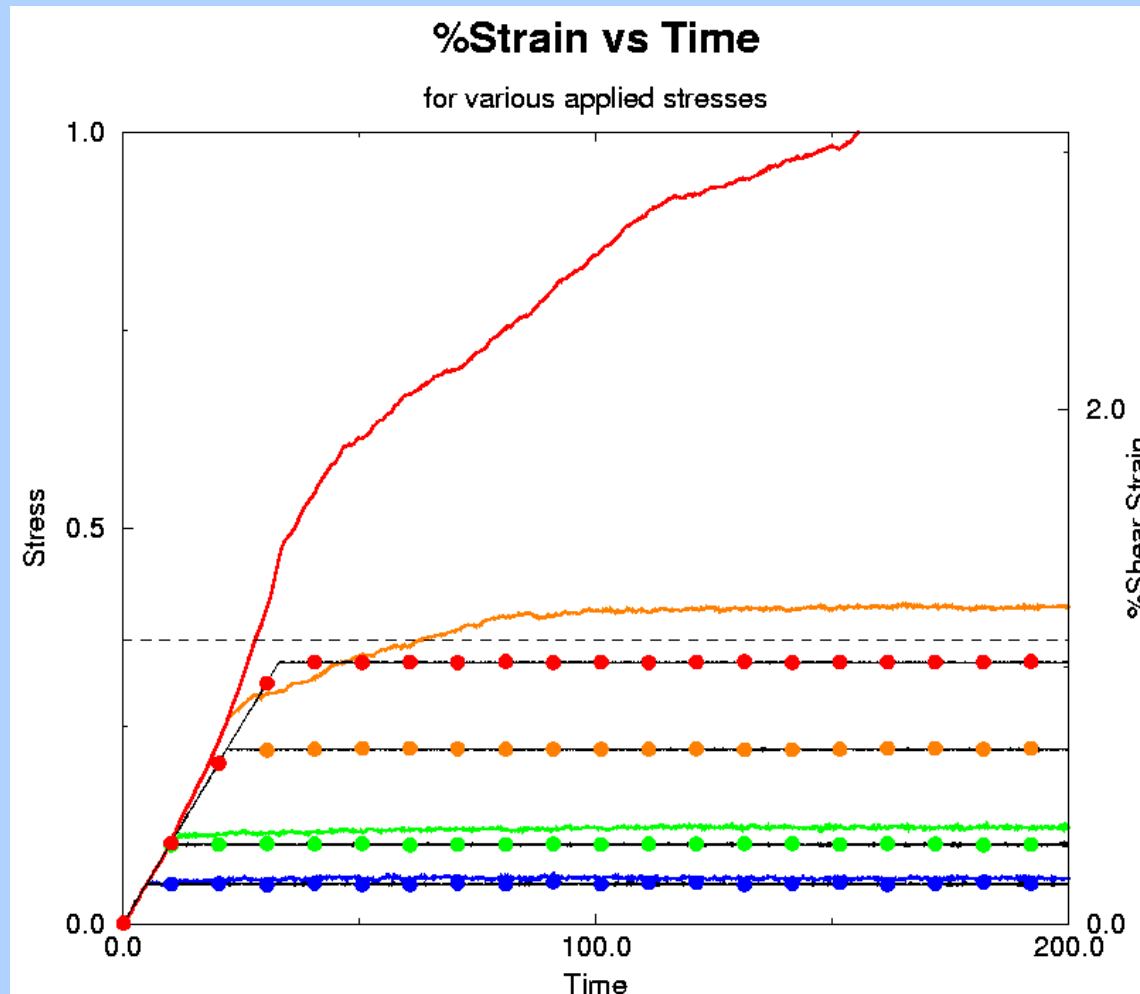
Widom, Strandburg, Swendsen (1987)

Homogeneous MD Simulations



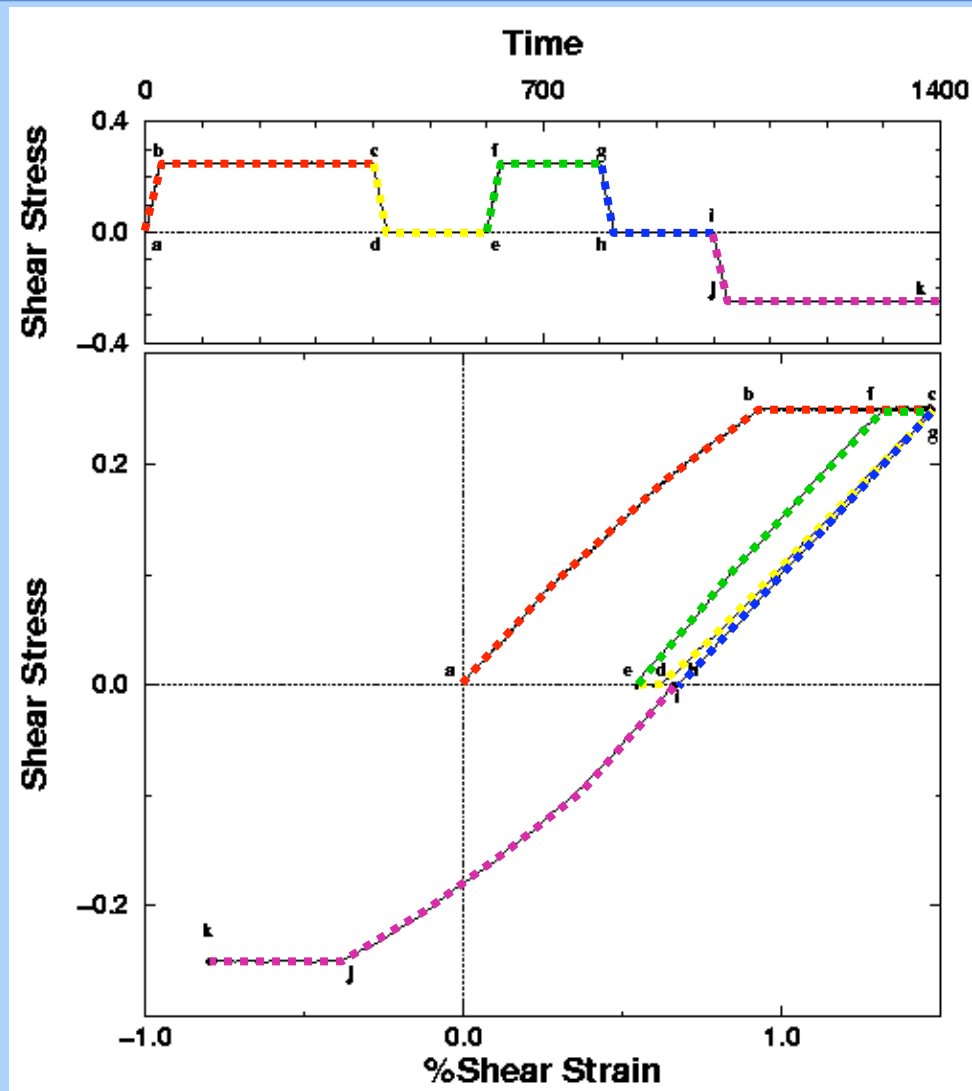
MLF and JS Langer, PRE, Vol. 57, pp. 7192-7205 (1998)

Homogeneous MD Simulations



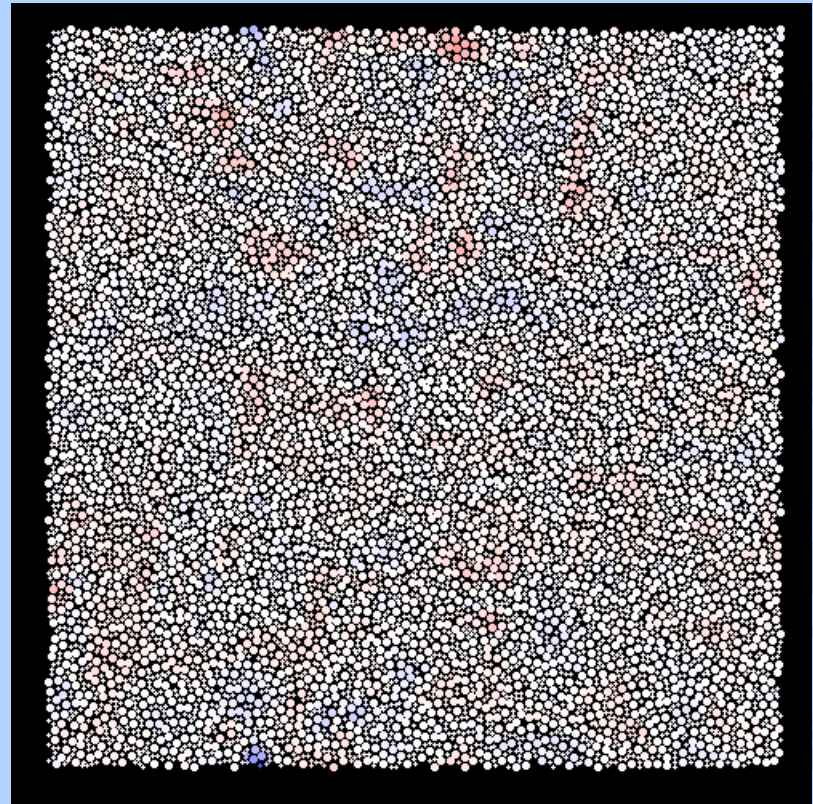
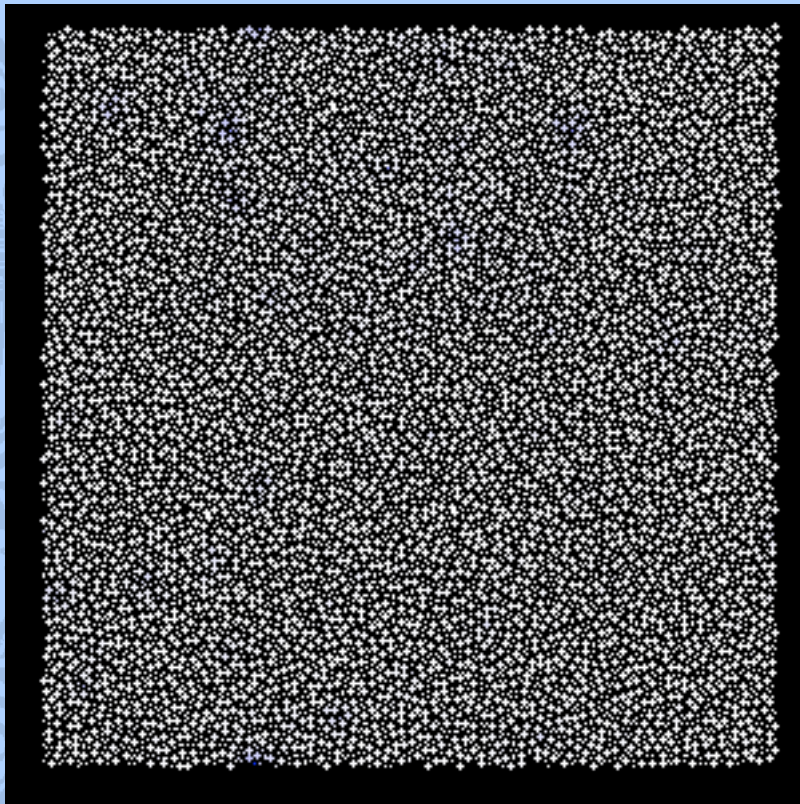
MLF and JS Langer, PRE, Vol. 57, pp. 7192-7205 (1998)

Homogeneous MD Simulations

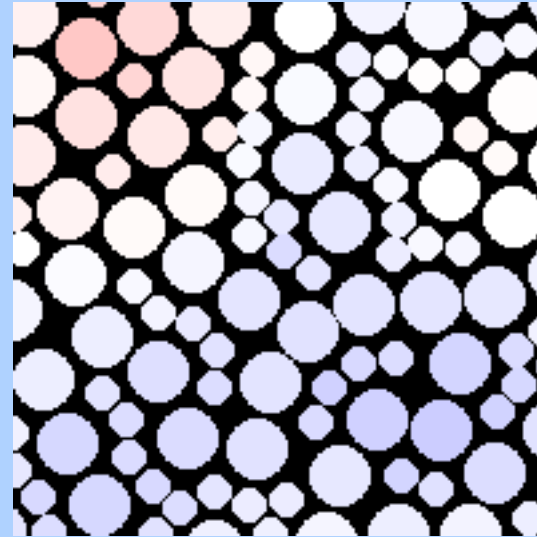
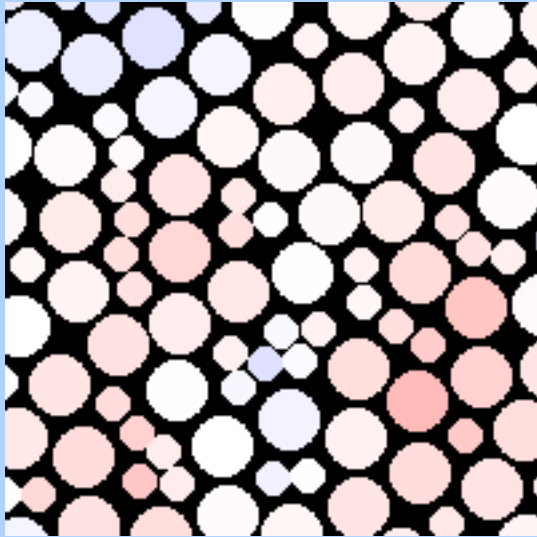


MLF and JS Langer, PRE, Vol. 57, pp. 7192-7205 (1998)

STZ Picture



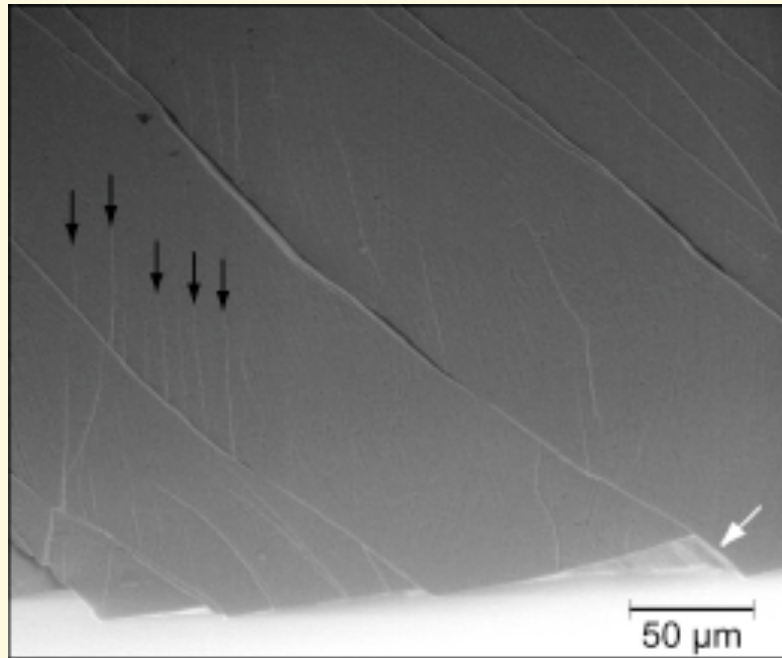
STZ Picture



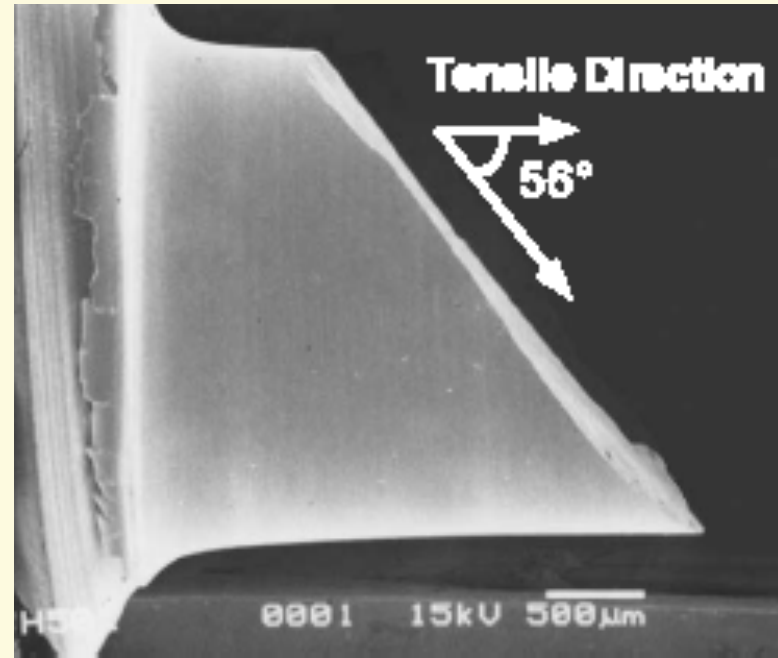
- **STZs have a particular orientation. They are susceptible to shear to the extent that the shear is along this direction.**
- **STZs are reversible until their environment rearranges. They behave as 2-state systems.**
- **STZs are transient. They can be created and destroyed by neighboring plastic activity.**

Metallic Glass Failure

strain localization (shear banding) is the primary failure mode

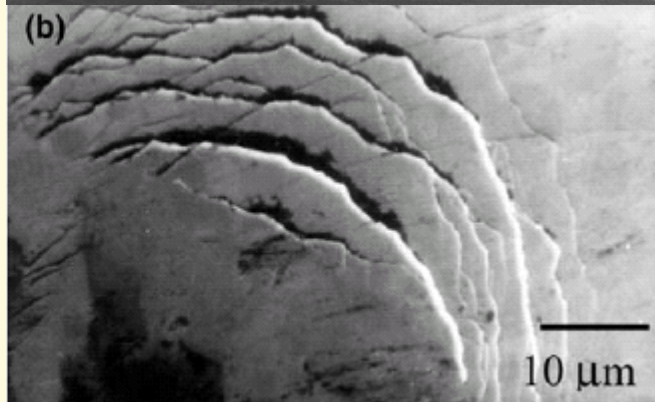
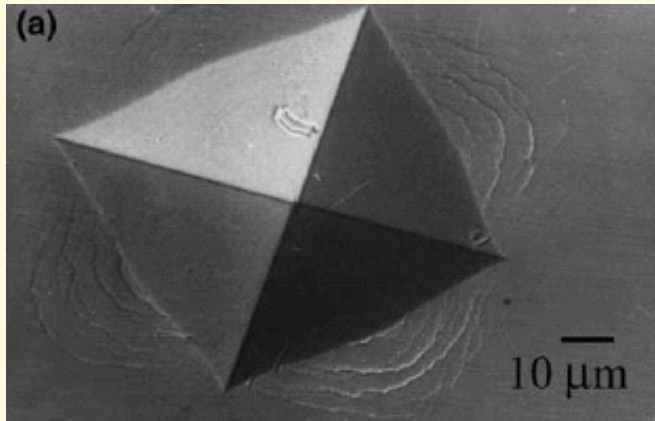


**Electron Micrograph of Shear Bands
Formed in Bending Metallic Glass**
Hufnagel, El-Deiry, Vinci (2000)



Quasistatic Fracture Specimen
Mukai, Nieh, Kawamura, Inoue,
Higashi (2002)

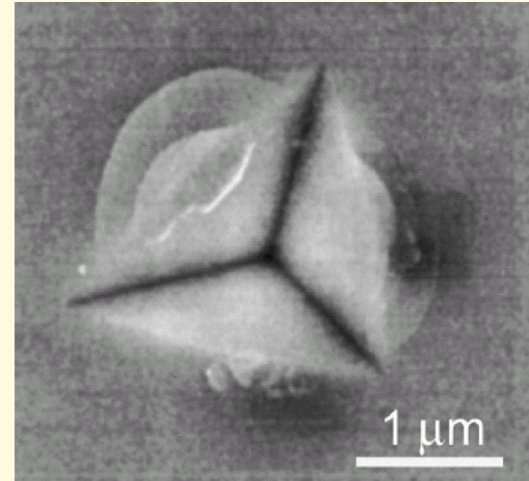
Indentation Testing of Metallic Glass



“Hardness and plastic deformation in a bulk metallic glass”

Acta Materialia (2005)

U. Ramamurty, S. Jana, Y. Kawamura, K. Chattopadhyay



“Nanoindentation studies of shear banding in fully amorphous and partially devitrified metallic alloys”

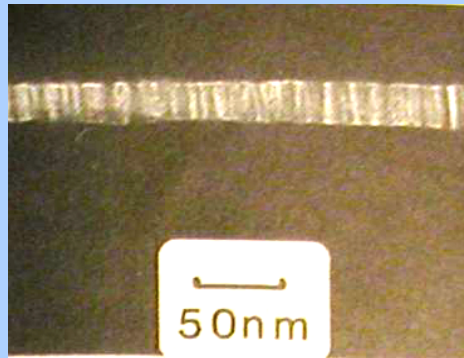
Mat. Sci. Eng. A (2005)

A.L. Greer., A. Castellero, S.V.

Madge, I.T. Walker, J.R. Wilde

Shear Bands

Polymer Craze



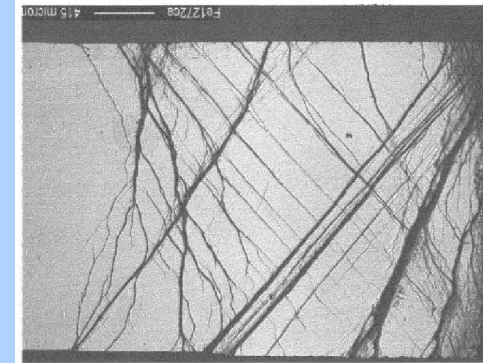
Young and Lovell (1991)

Mild Steel



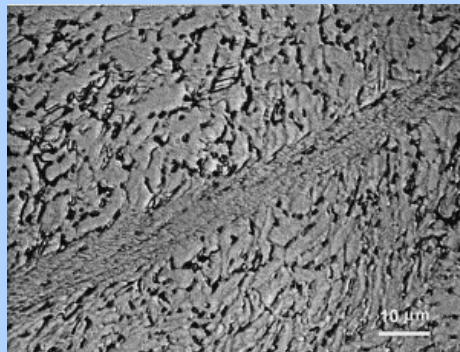
Van Rooyen (1970)

Nanograined Metal



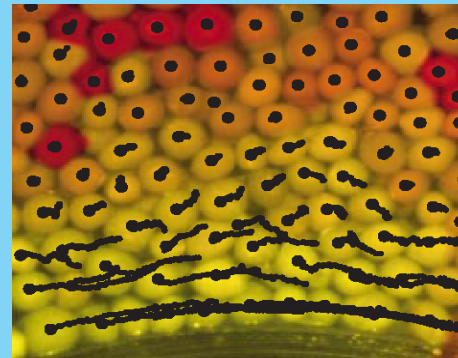
Wei, Jia, Ramesh and Ma (2002)

Steel @ High Rate



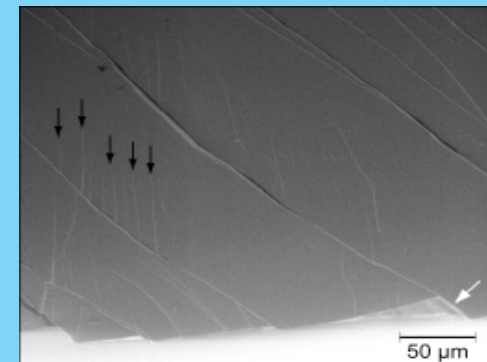
Xue, Meyers and Nesterenko (1991)

Granular Materials



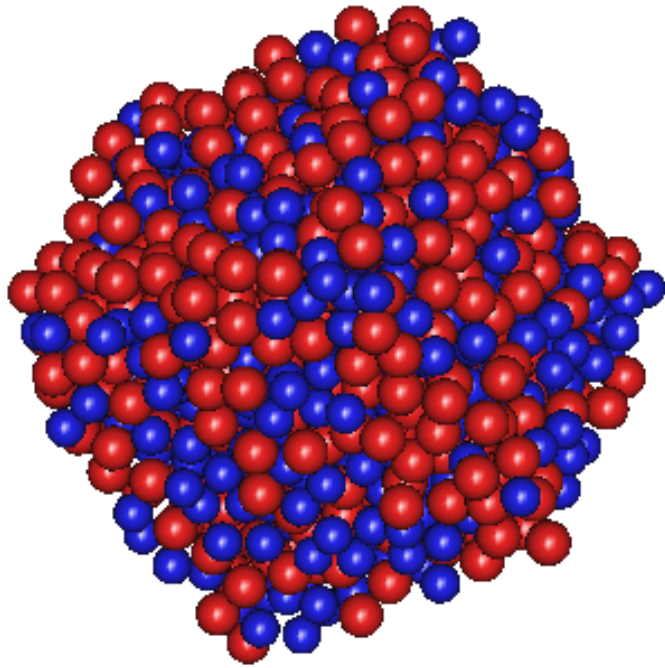
Mueth, Debregeas and et. al. (2000)

Bulk Metallic Glasses



Hufnagel, El-Deiry and Vinci (2000)

Simulated System: 3D Binary Alloy

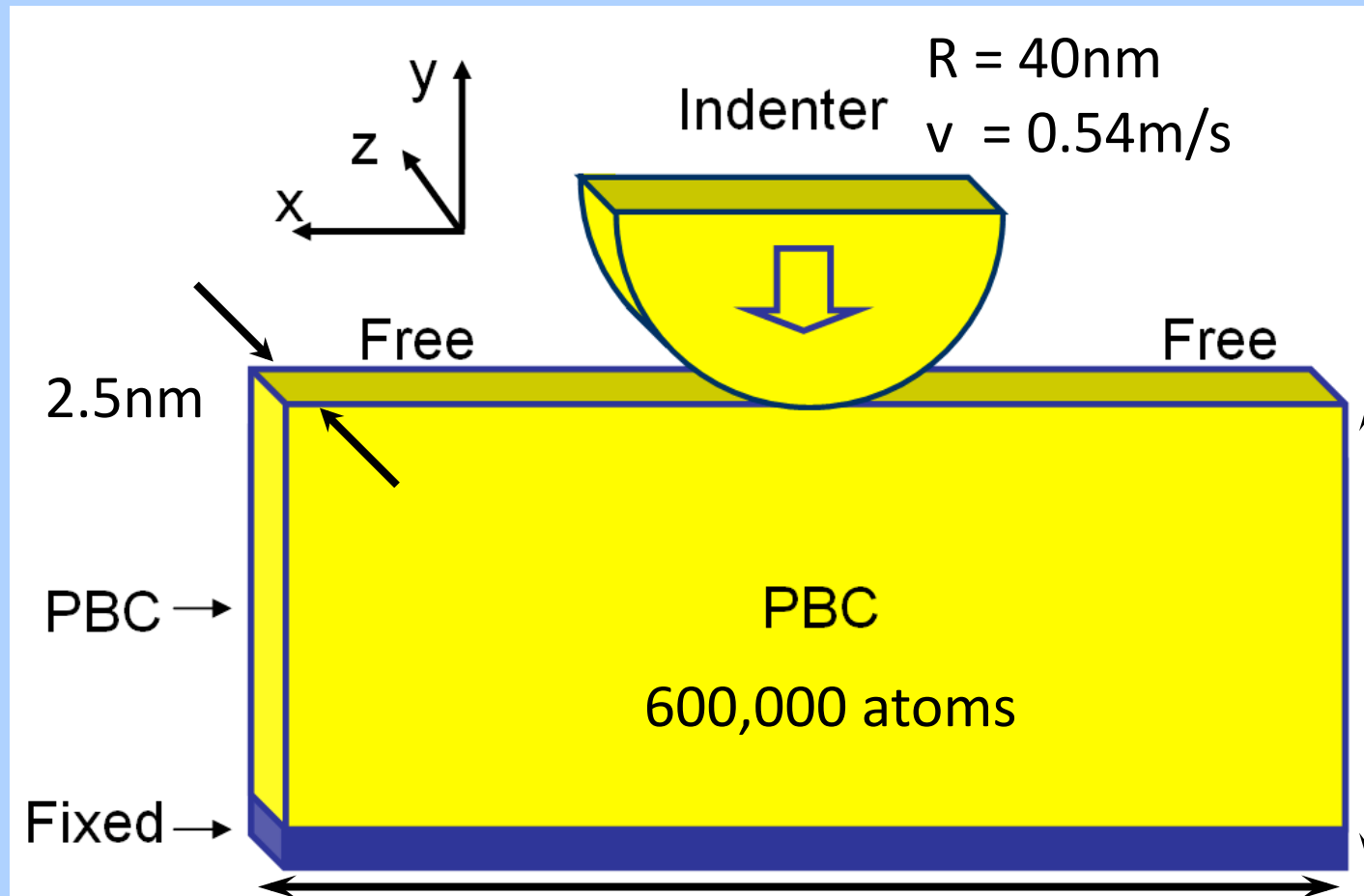


- Wahnstrom Potential (PRA, 1991)
- Rough Approximation of $\text{Nb}_{50}\text{Ni}_{50}$
- Lennard-Jones Interactions
- Equal Interaction Energies
- Bond Length Ratios:
 - $a_{\text{NiNi}} \sim \frac{5}{6} a_{\text{NbNb}}$
 - $a_{\text{NiNb}} \sim \frac{11}{12} a_{\text{NbNb}}$
- $T_g \sim 1000\text{K}$
- Studied previously in the context of the glass transition (Lacevic, *et. al.* PRB 2002)

- Unlike the simulation of crystalline systems, it is not possible to skip simulating the processing step
- Glasses were created by quenching at 3 different rates: 50K/ps, 1K/ps and 0.02 K/ps

Metallic Glass Nanoindentation

Simulations performed using molecular dynamics code across 64 nodes of a parallel cluster

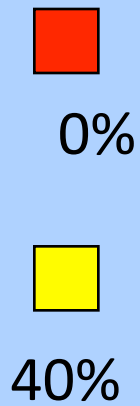


Y. Shi, MLF, Acta Materialia, 55, 4317 (2007)

100nm

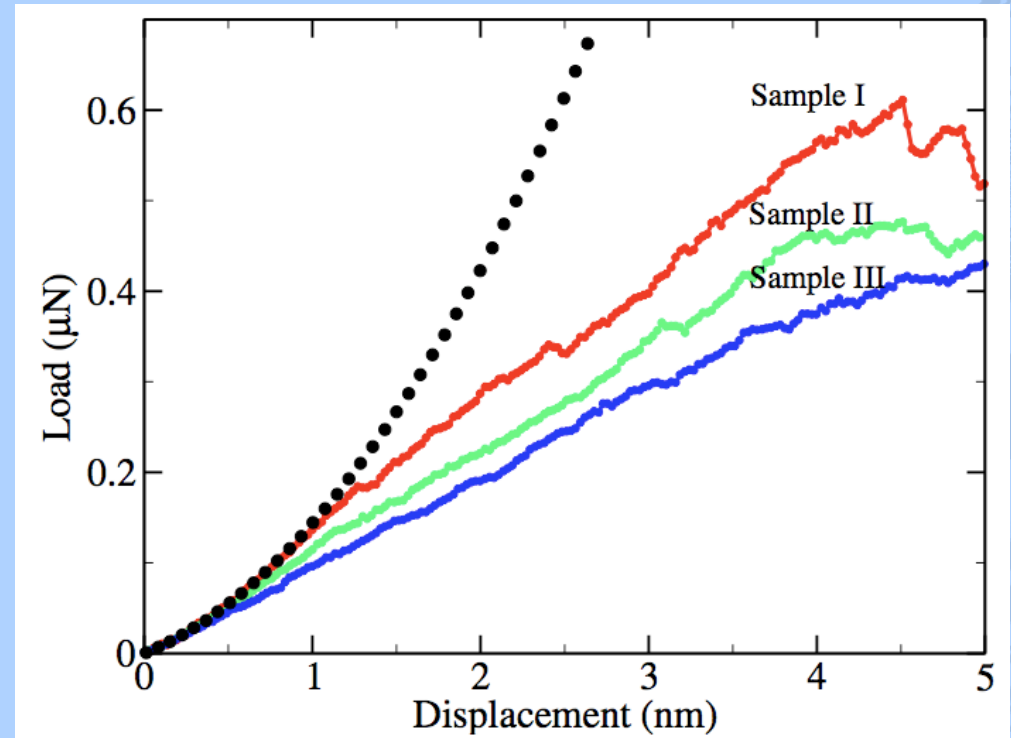
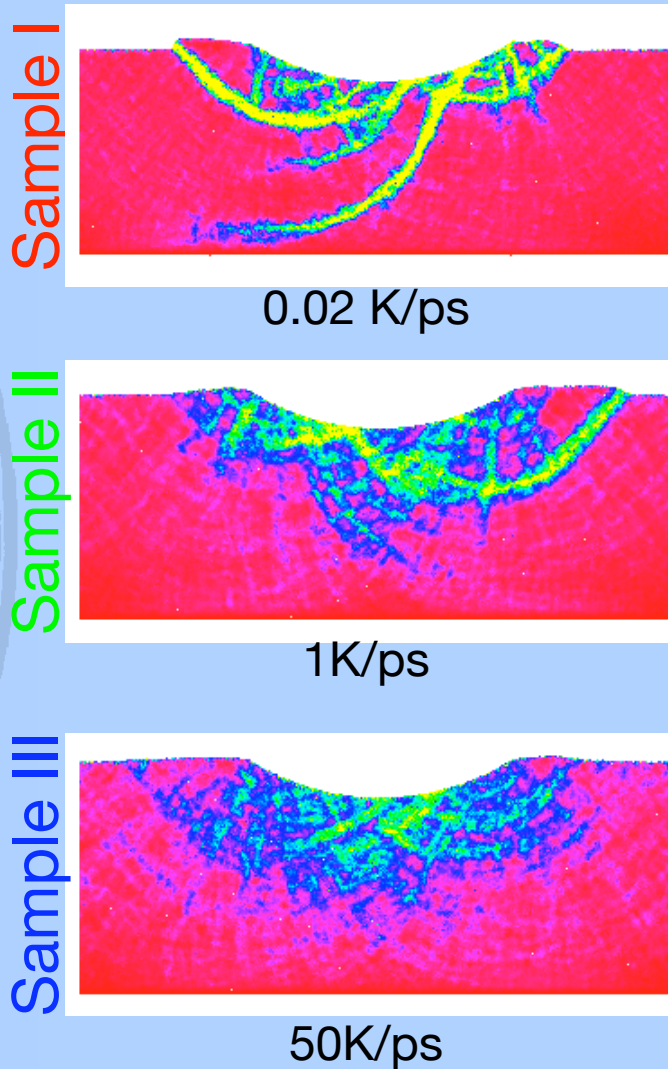
Metallic Glass Nanoindentation

color = deviatoric strain



Y. Shi, MLF, Acta Materialia, 55, 4317 (2007)

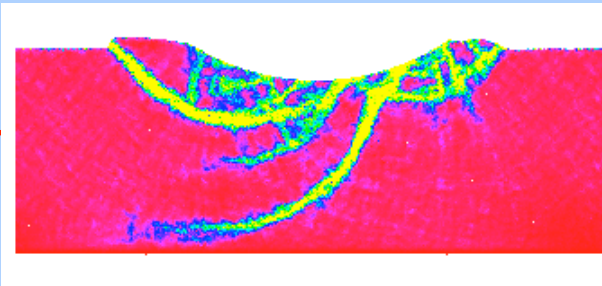
Metallic Glass Nanoindentation



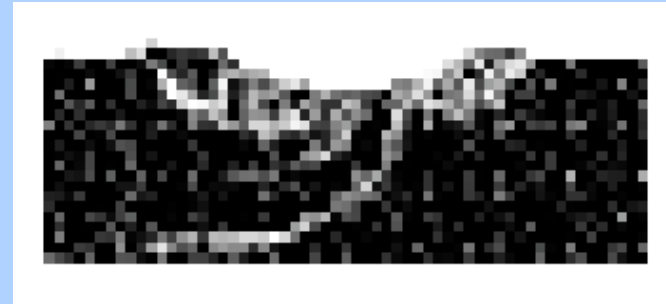
Y. Shi, MLF, Acta Materialia, 55, 4317 (2007)

Metallic Glass Nanoindentation

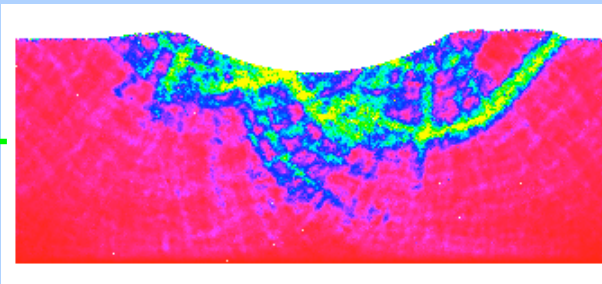
Sample I



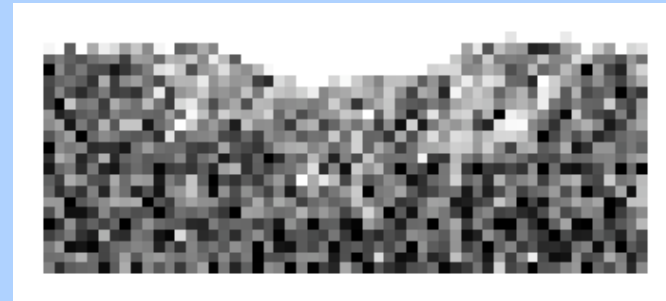
0.02 K/ps



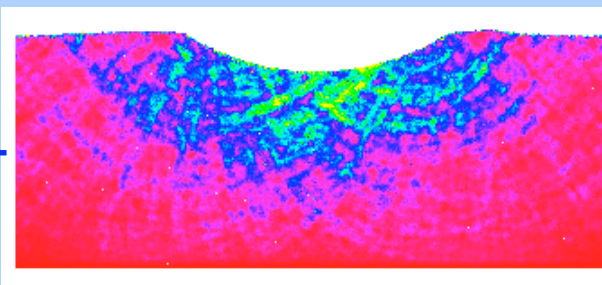
Sample II



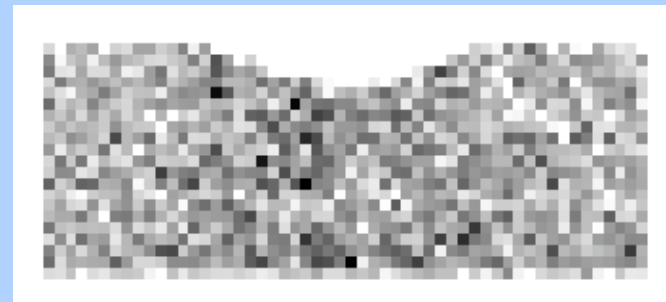
1K/ps



Sample III



50K/ps



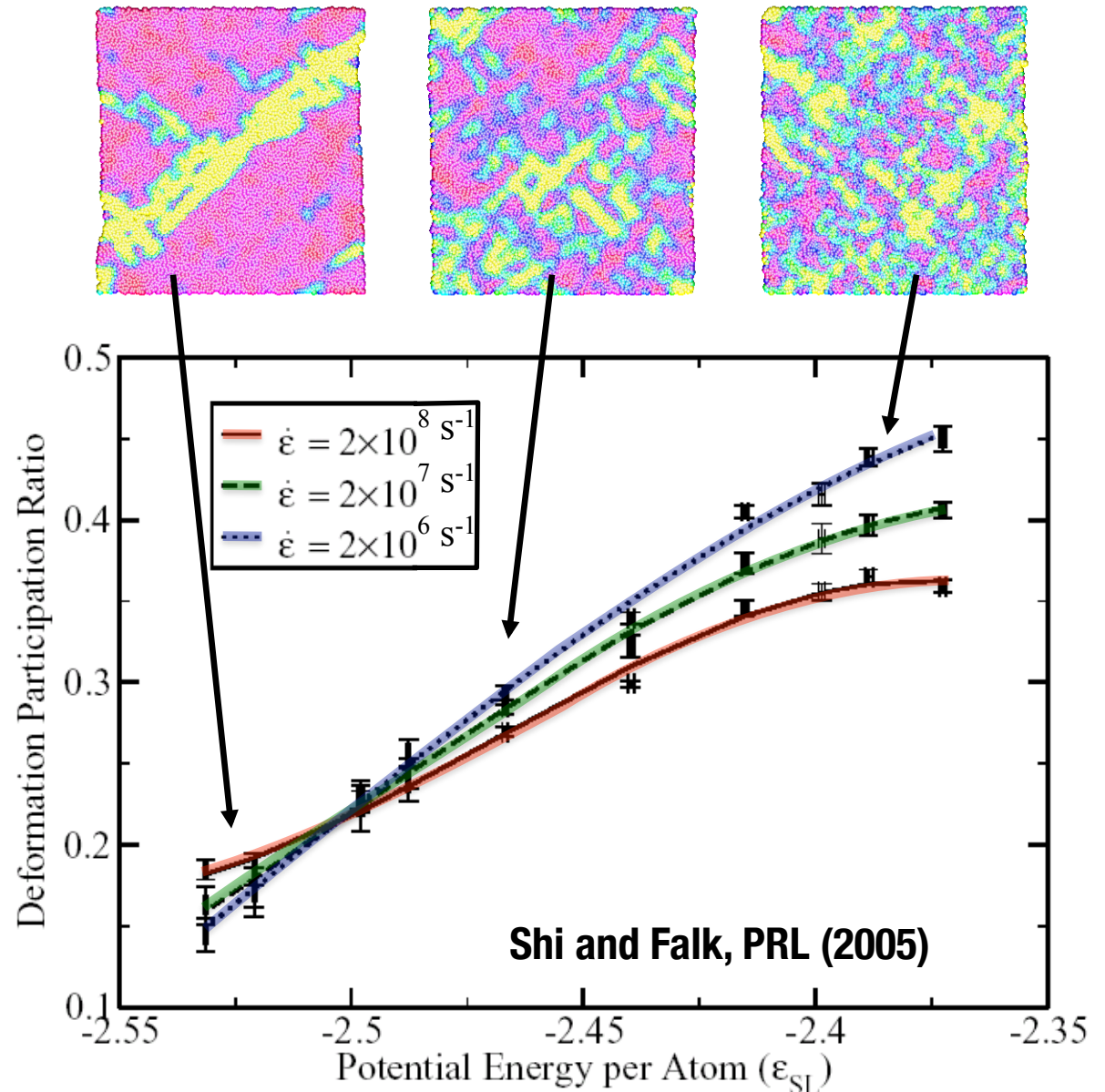
Y. Shi, MLF, Acta Materialia, 55, 4317 (2007)

Quantifying the Dependence of Localization on Quench Rate (2D)

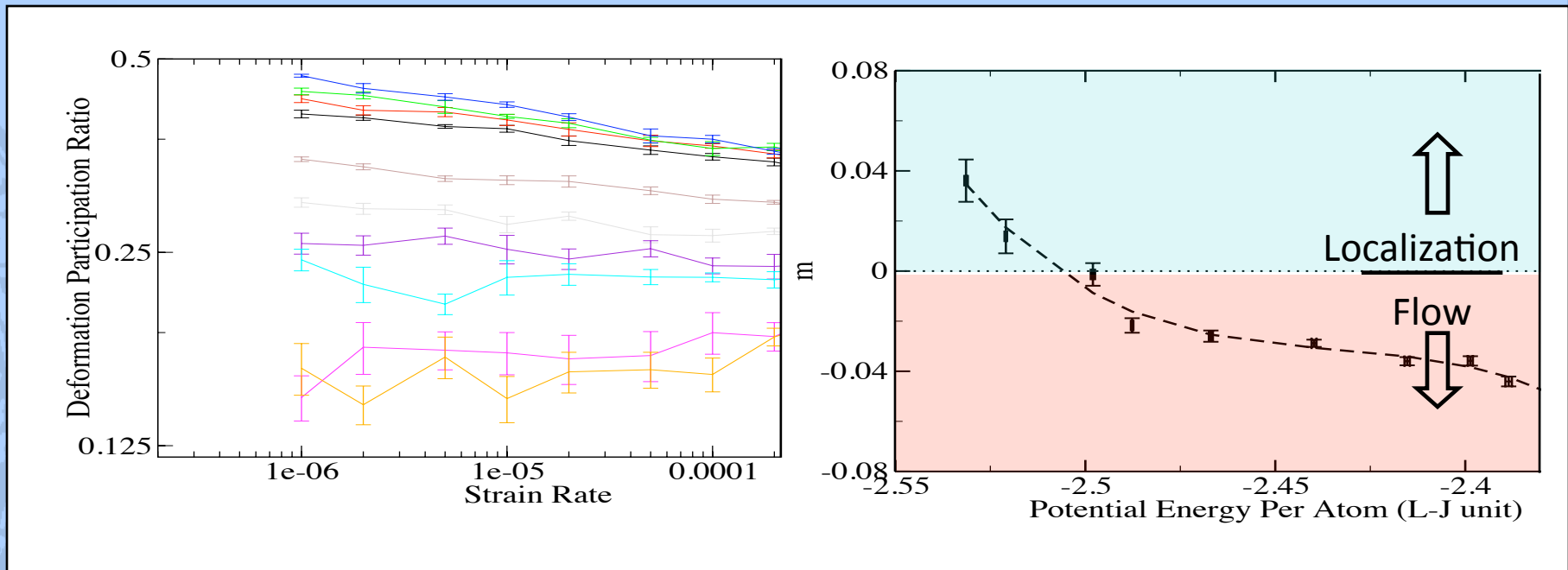
- Performed **756** individual 2D uniaxial tensile test simulations at $0.1 T_g$
- **10 different quench schedules** starting from equilibrium liquids
- **6-10 samples** at each quench schedule
- Each of these 84 specimens was tested at **9 different strain rates** spanning 2 orders of magnitude

Shear Localization (DPR) vs. PE

- **Deformation Participation Ratio:** Percentage of material with a local shear strain larger than the nominal strain
- Low strain rate favors homogenous deformation in instantaneously quenched samples
- Low strain rate favors inhomogeneous deformation in gradually quenched samples.



Strain-rate sensitivity of DPR



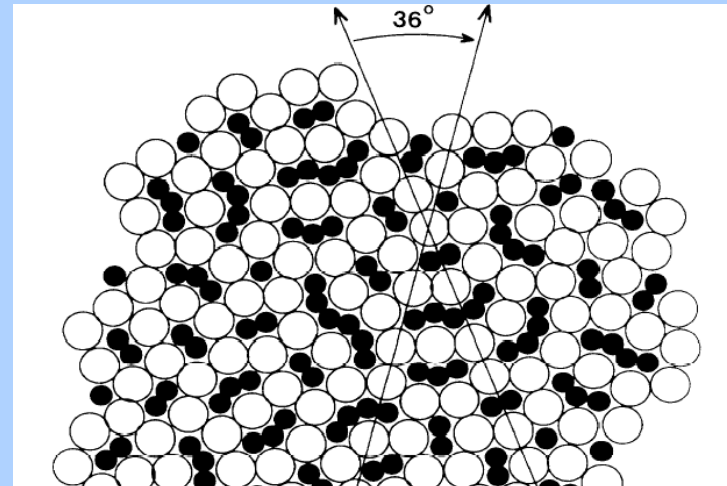
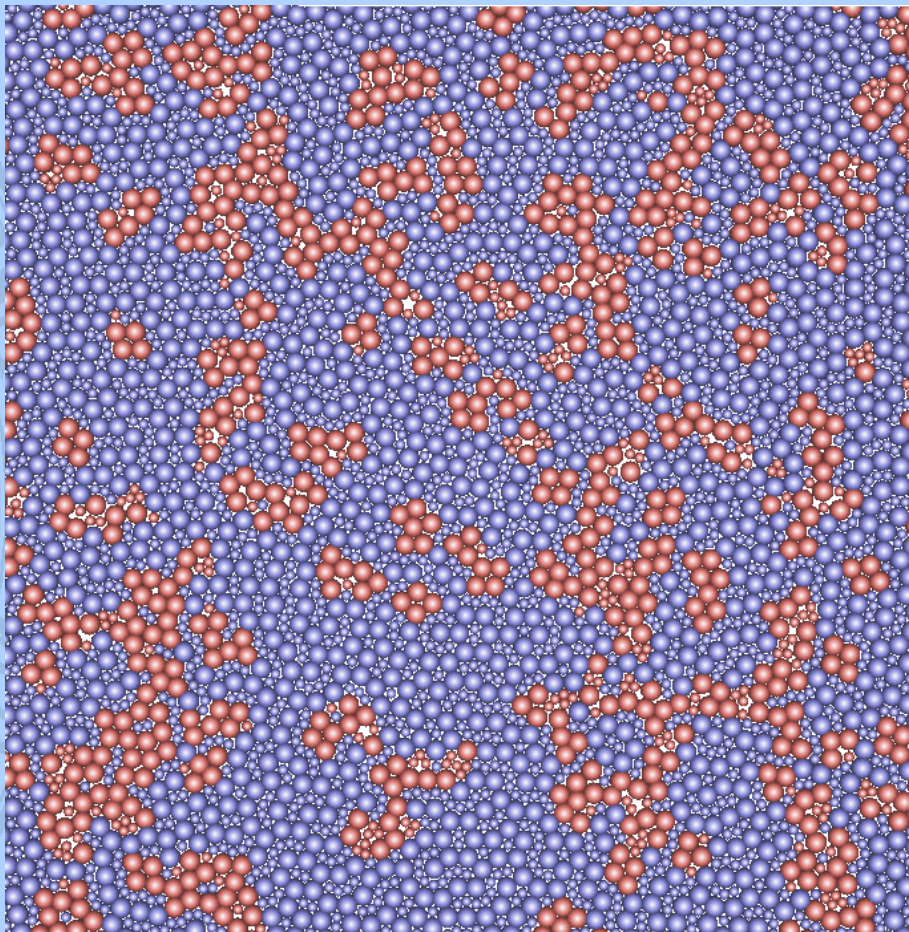
$$DPR \approx A \dot{\epsilon}^m$$

For $\dot{\epsilon} \rightarrow 0$ and system size $\rightarrow \infty$

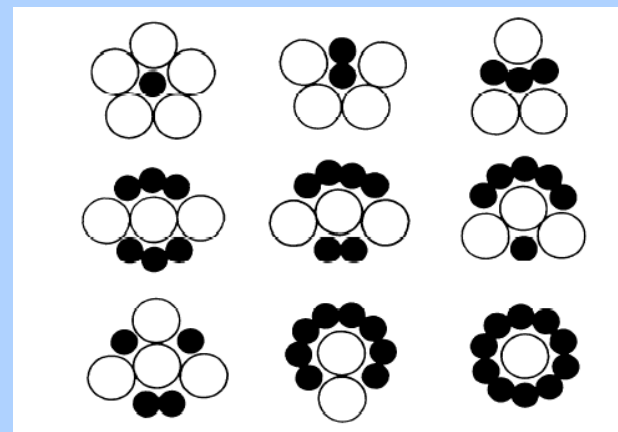
$m < 0$: homogenous deformation

$m \geq 0$: localized deformation

Local Structural Analysis

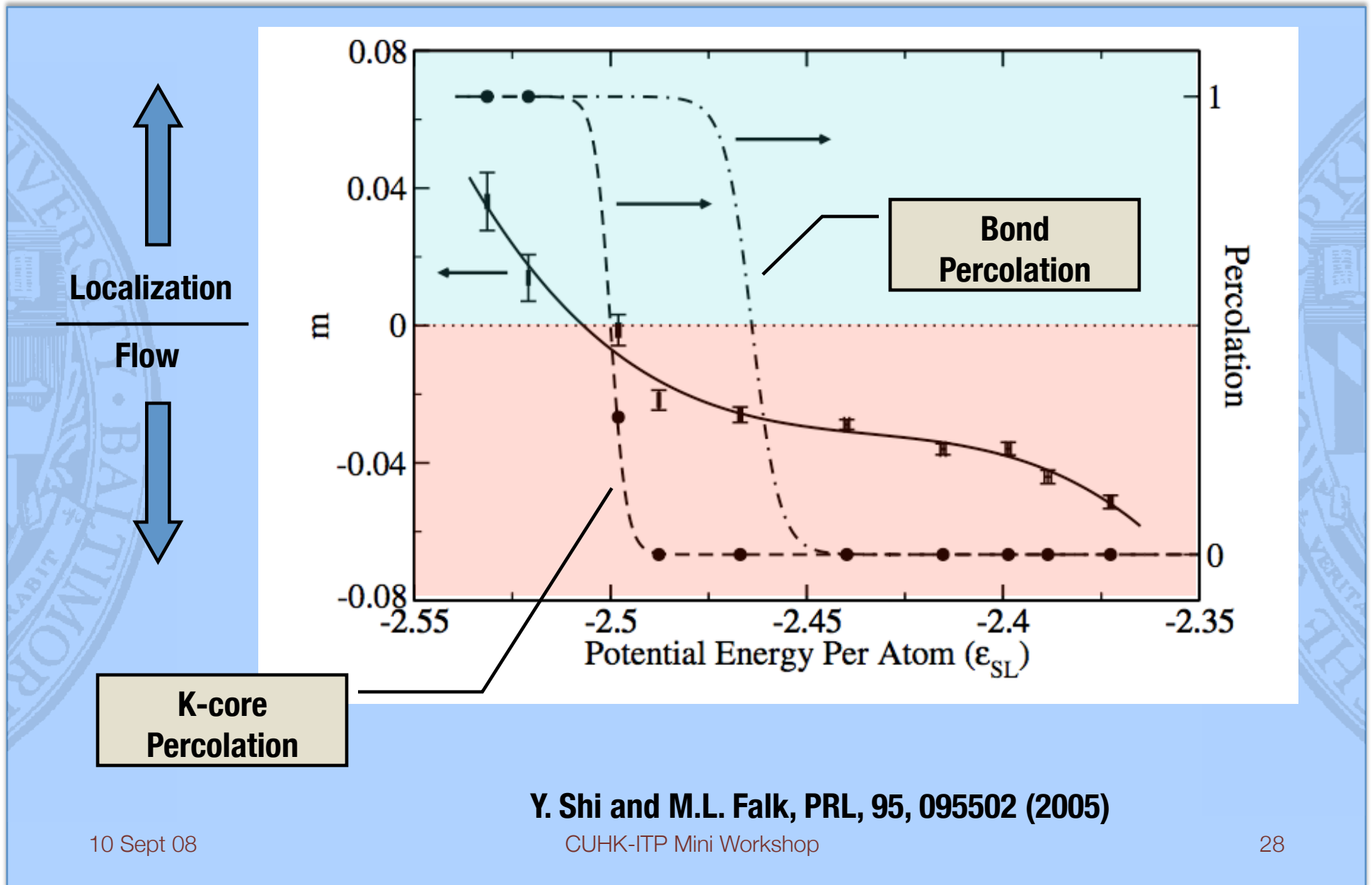


Widom, Strandburg, Swendsen (1987)



Complete set of low-energy local environments (Widom, 1987)

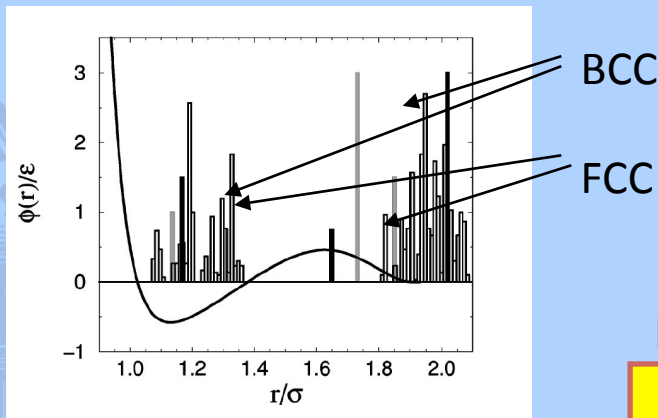
Percolation and Localization



Y. Shi and M.L. Falk, PRL, 95, 095502 (2005)

3D Simulation Potentials

Dzugutov Potential

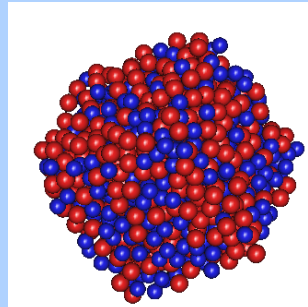


Roth and Denton, PRE (2000)

- 3D Monoatomic
- Energy penalties for crystalline phases
- Dodecagonal quasicrystal
- $T_{MCT} = 0.4$

Zetterling et al., JNCS (2001)

Wahnstrom LJ Binary



Bond length Bond strength

AA 1.000
AB 0.917
BB 0.833

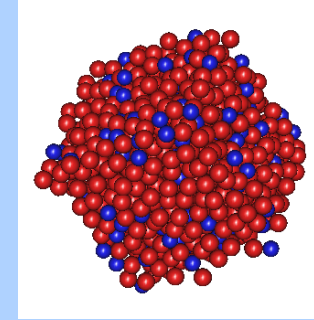
AA 1.0
AB 1.0
BB 1.0

- 3D binary LJ 12-6 potential
- 50:50 composition, 144,000 atoms
- $T_{MCT} = 0.57$

Wahnstrom, PRA, 1991

Lacevic et al., PRB, 2002

Kob-Andersen LJ Binary



Bond length Bond strength

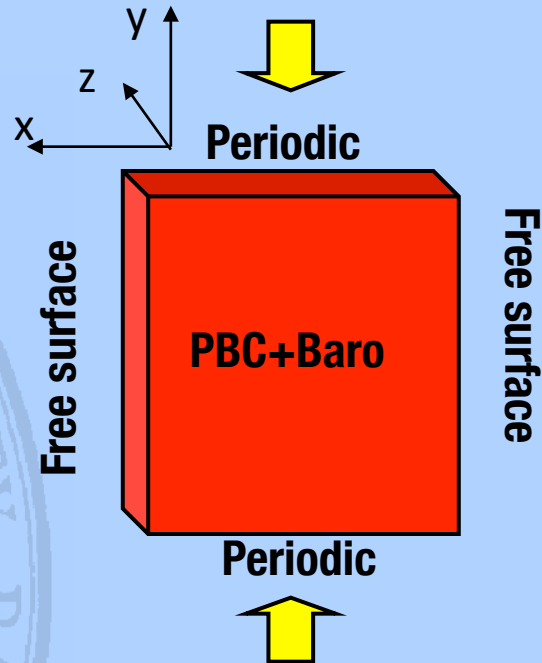
AA 1.00
AB 0.80
BB 0.88

AA 1.0
AB 1.5
BB 1.0

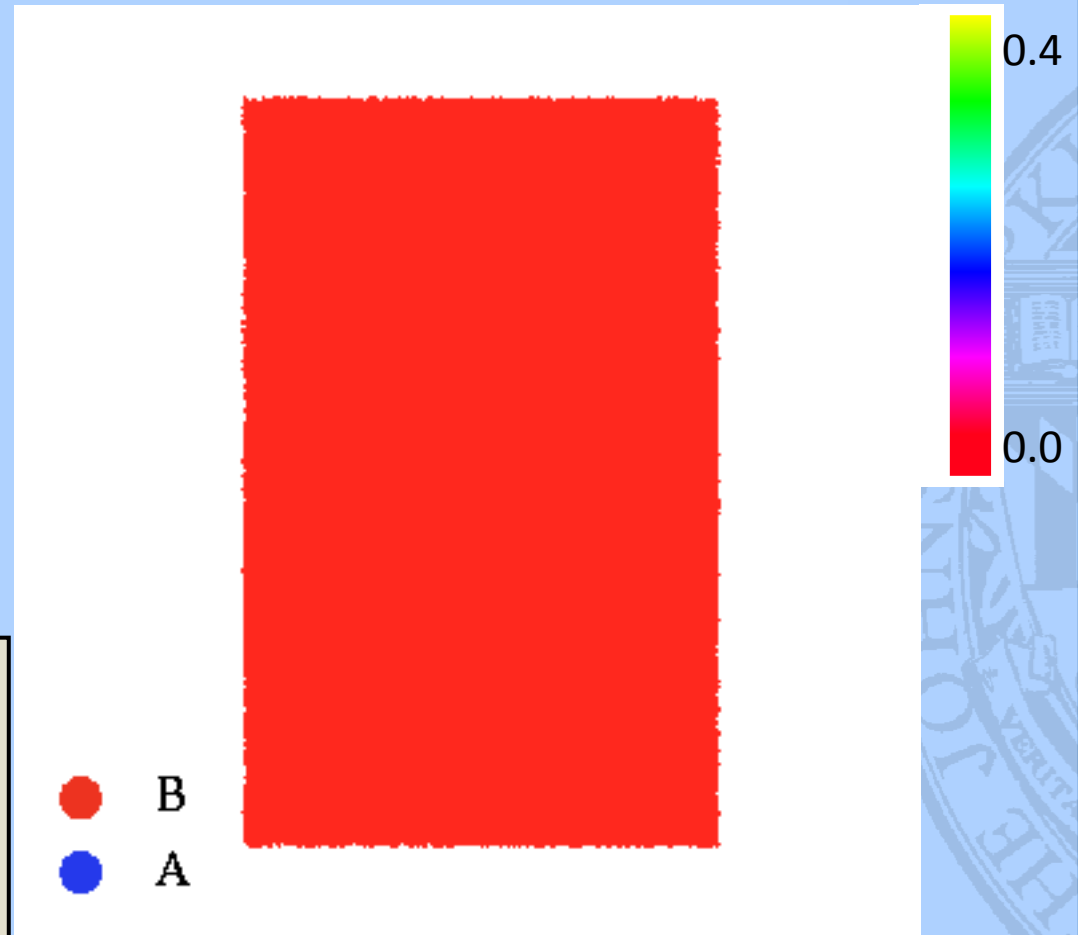
- 3D binary LJ 12-6 potential
- 80:20 composition, 144,000 atoms
- $T_{MCT} = 0.435$

Kob and Andersen, PRE 1995

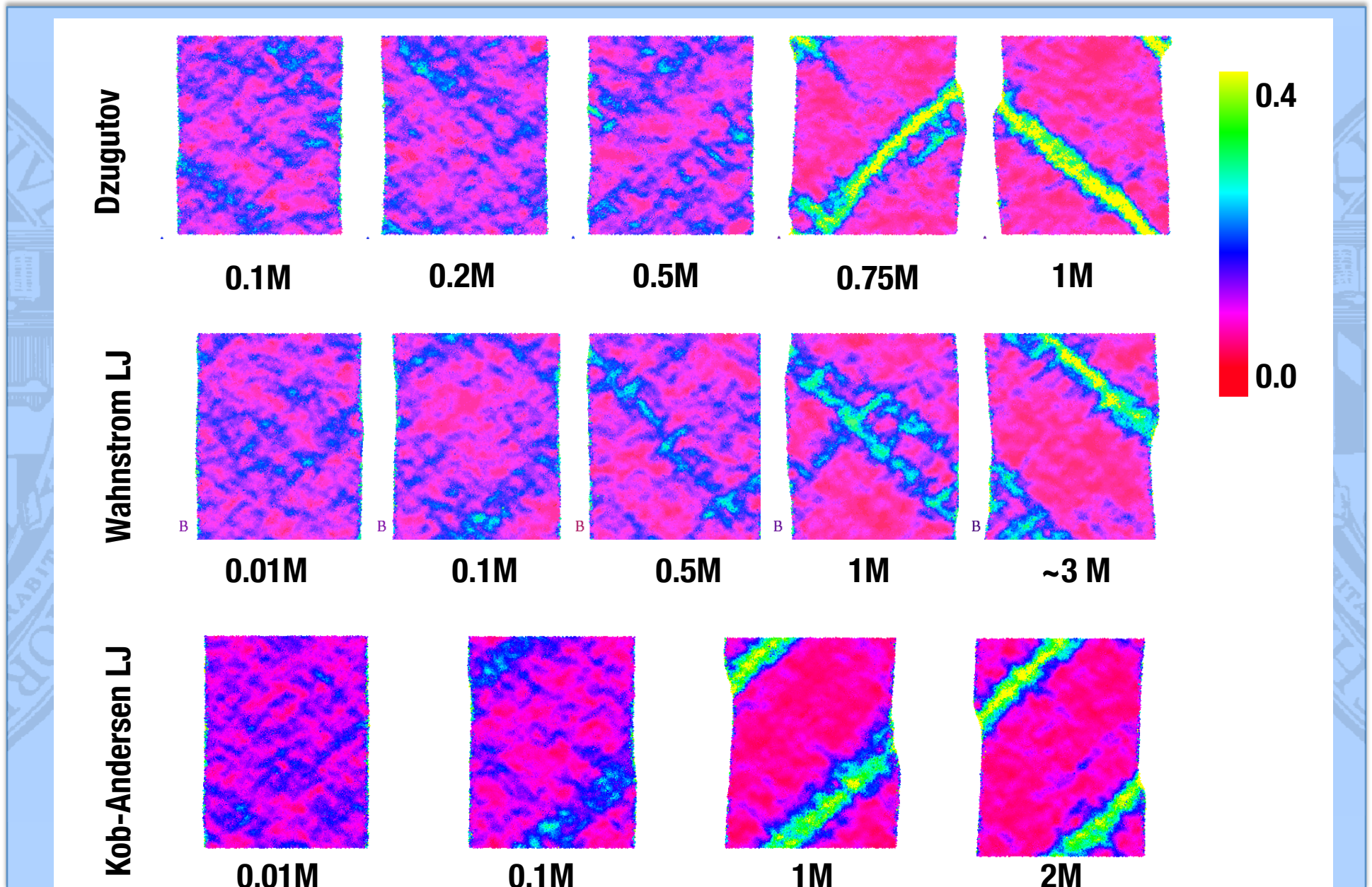
3D Uniaxial Compression Test



- Thin slab geometry to maximize in-plane spatial dimension
75×110 ×15: 140,000 atoms
- Free surfaces in Y-Z
- PBC in X-Y and Y-Z
- Plane Strain: Average σ_{zz} zero

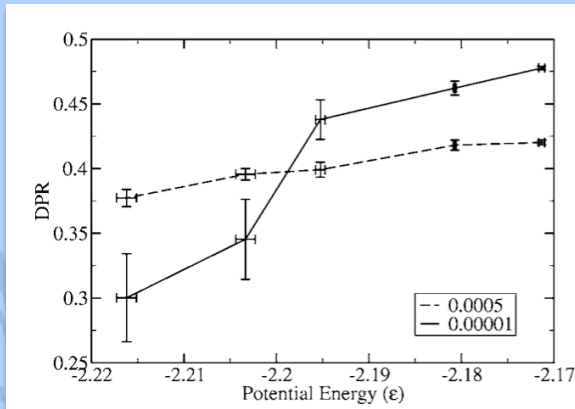


Compression: Various Quenches



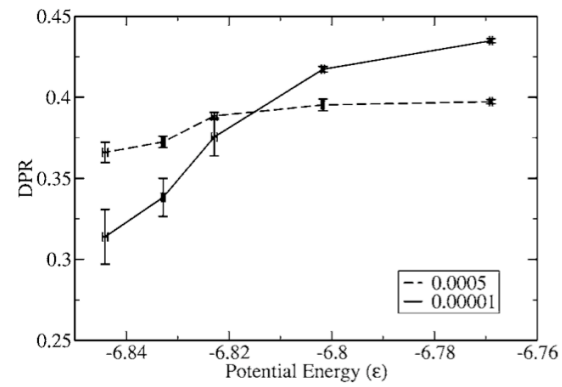
DPR and Strain Rate Sensitivity

Dzugutov System



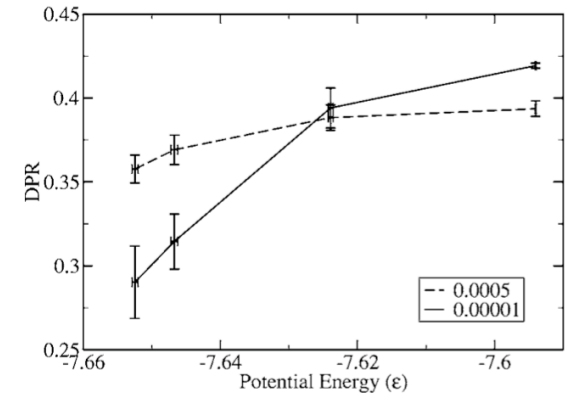
PE

Wahnstrom LJ



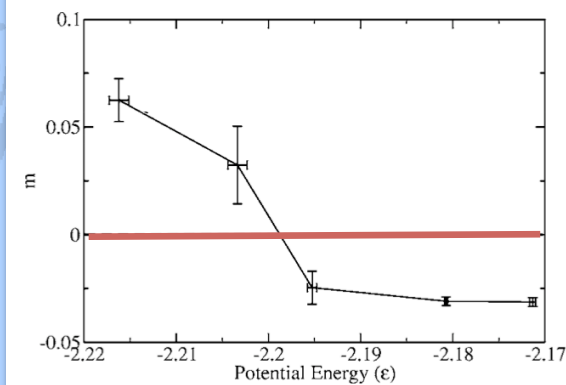
PE

Kob-Andersen LJ

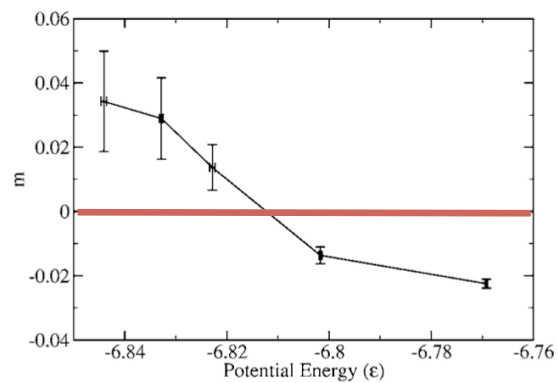


PE

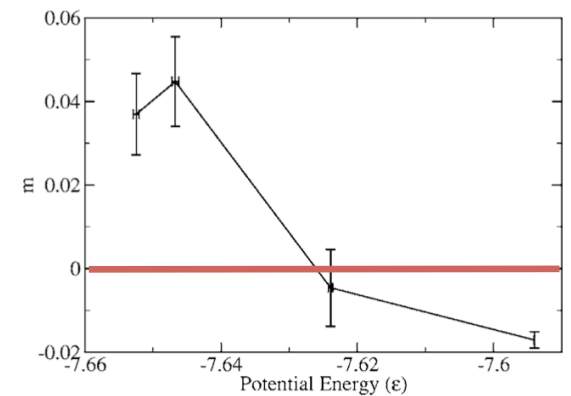
m



PE



PE



PE

Triangulated Coordination Shell Analysis

Triangulated Coordination Shells: Bonds by atoms within the coordination shell form only triangles. The center atom and the triangle has to form a space dividing tetrahedron.

Criterion: (From Euler's formula)

$$\sum_q (6 - q)v_q = 12$$

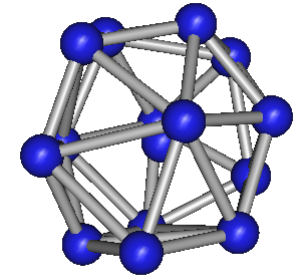
q is the surface coordination number (from 3 to 8 for now)

v_q is the count of neighbors has surface coordination number q

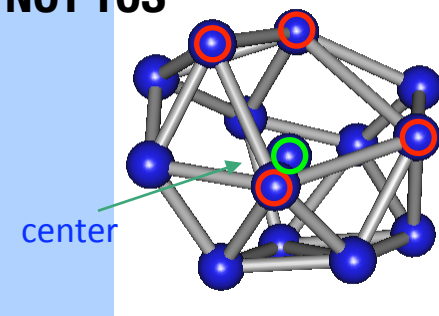
Glassy samples with lowest quenching rate

	TCS	Icosahedra
Dzugutov	25 %	12 %
Wahnstrom	13 %	10 %
K-A	3%	0.1 %

TCS

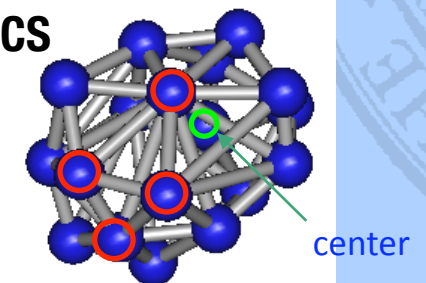


NOT TCS



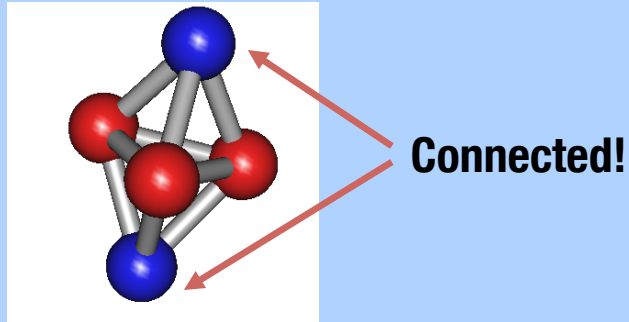
The four red atoms are forming a non-planar quadrilateral not a triangle

NOT TCS



The tetrahedron formed by 4 red atoms are space dividing but not consisting the center atom (green)

3D Percolation Analysis



Two atoms sharing at least three atoms are “connected”

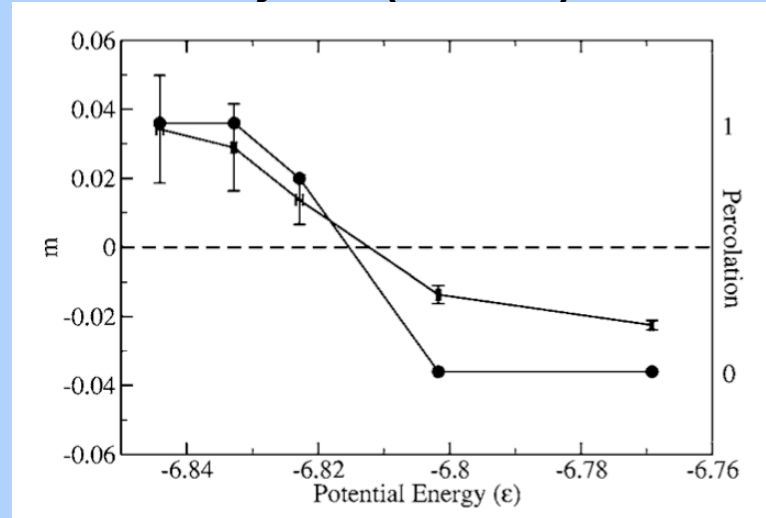
The cage of two atoms have to interpenetrate or sharing faces

Similar to Zettering, et al, JNCS, 2001

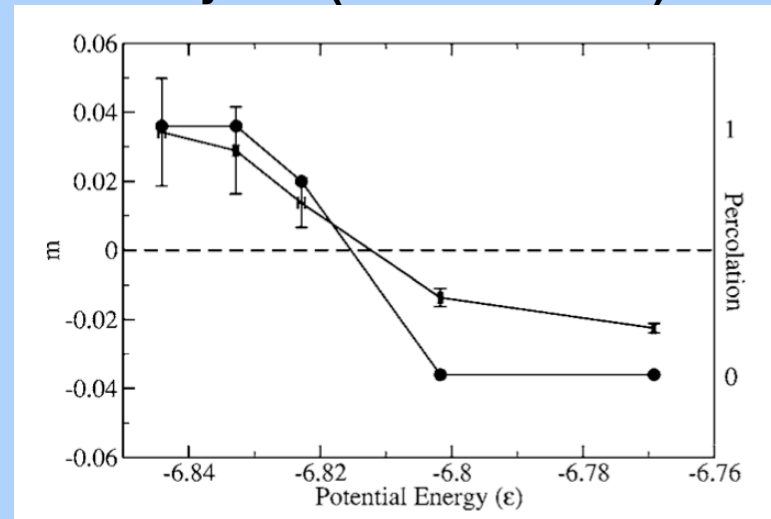
DZ: (all percolate)

KA: (none percolate)

WA system (TCS SRO)



DZ system (Icosahedral SRO)



Observations

- **Experimental investigations of metallic glass reveal interesting behavior:**
 - Shear softening
 - Nonlinear Rheology
 - Systematic changes in ductility and creep dependent on alloy composition and sample history
 - Strong tendency to localization
- **Simulations confirm these and also reveal**
 - Plastic Hysteresis (Bauschinger effects)
 - Existence of a yield stress
 - Local nature of rearrangements
- **Recent investigations of shear banding show**
 - Slowly quenched samples more susceptible to localization
 - Samples that exhibit localization contain percolated short range order
 - CAVEAT: there is no universal method for defining short range order
- **Tomorrow: building a theory of plasticity**