

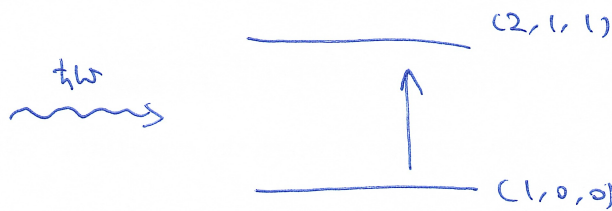
SQ 25

In SQ 24, we have already discussed the transition from

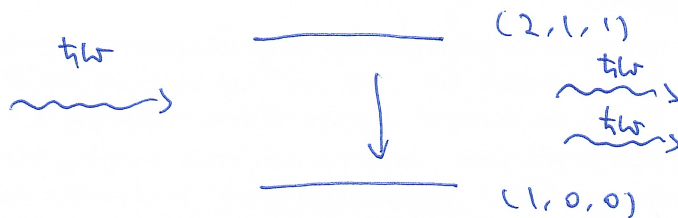
$$\begin{array}{ccc}
 2p & \longleftrightarrow & 1s \\
 (n=2, l=1, m_l=\pm 1) & & (n=1, l=0, m_l=0)
 \end{array}$$

There are 3 possible ways of transitions:

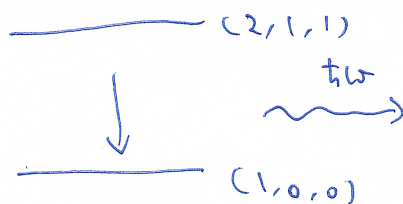
① Stimulated Absorption ($1s \rightarrow 2p$), B_{12}



② Stimulated Emission ($2p \rightarrow 1s$), B_{21}



③ Spontaneous Emission ($2p \rightarrow 1s$), A



In SQ 24, we have discussed process ①, and we obtained

$$a_{(2,1,1)}(t) \propto \vec{r}_{(2,1,1), (1,0,0)} = \frac{\sqrt{2}}{243} a_0 (x^2 - iy^2)$$

(Reminder: $\vec{r}_{(2,1,1), (1,0,0)}$ is a complex vector)

a. Next, we move to process ②.

An important note is: Both process ① and ② have the same perturbation, a stimulation of photons " $\overset{h\nu}{\rightsquigarrow}$ ".

Therefore, the perturbation term of the Hamiltonian should be the same, i.e.,

$$\hat{H}' = -\vec{\mu}_{el} \cdot \vec{E} = \vec{r} \cdot (e\vec{E}_0 \cos \omega t)$$

∴ We can write down the integral of $\bar{\Gamma}_{1s,2p}$ to be:

$$\bar{\Gamma}_{1s,2p} = \int \psi_{1s}^*(\vec{r}) \vec{r} \psi_{2p}(\vec{r}) d^3r$$

Taking the complex conjugate,

$$\bar{\Gamma}_{1s,2p}^* = \int [\psi_{1s}^*(\vec{r})]^* \vec{r}^* \psi_{2p}^*(\vec{r}) [d^3r]^*$$

Since the components \vec{r} and d^3r are real,

$$\begin{aligned} \bar{\Gamma}_{1s,2p}^* &= \int \psi_{1s}(\vec{r}) \vec{r} \psi_{2p}^*(\vec{r}) d^3r \\ &= \int \psi_{2p}^*(\vec{r}) \vec{r} \psi_{1s}(\vec{r}) d^3r \\ &= \bar{\Gamma}_{2p,1s} \end{aligned}$$

Plug the result we obtained in EQ 24,

$$\begin{aligned} \bar{\Gamma}_{1s,2p} &= \bar{\Gamma}_{2p,1s}^* \\ &= \frac{128}{243} Q_0 (x^1 + iy^1) \end{aligned}$$

Note: $\bar{\Gamma}_{1s,2p}$ is also a complex vector.

b. To get the transition coefficients B_{21} , we need $|\bar{r}_{1s,2p}|^2$,

$$\begin{aligned} |\bar{r}_{1s,2p}|^2 &= \bar{r}_{1s,2p}^* \cdot \bar{r}_{1s,2p} \\ &= \left(\frac{128}{243} a_0\right)^2 (\hat{x} - i\hat{y}) \cdot (\hat{x} + i\hat{y}) \\ &= 2 \left(\frac{128}{243} a_0\right)^2 \\ &\approx 0.555 a_0^2 \end{aligned}$$

Note: $|\bar{r}_{2p,1s}|^2 = \left(\frac{128}{243} a_0\right)^2 (\hat{x} + i\hat{y}) \cdot (\hat{x} - i\hat{y})$
 $= |\bar{r}_{1s,2p}|^2$

\therefore The transition coefficients B_{12} (Process ①) is actually the same as B_{21} (Process ②).

Some background to be optional:

Now, we have already known that

$$B \propto |\bar{r}_{12}|^2,$$

Using the fact that:

In thermal equilibrium, the number of atoms in state 1s and state 2p are maintained such that the rate of "①, stimulated absorption" is the same as the rate of "②, stimulated emission" plus "③, spontaneous emission".

So this is a link between A coefficient to B coefficient.

c. Finally, we get $A = \frac{\omega^3}{3\pi\epsilon_0 c^3 \hbar} e^2 |\bar{r}_{21}|^2$,

which depends on 3 factors, a bunch of constants $\frac{1}{3\pi\epsilon_0 c^3 \hbar}$,

ω^3 dependences, and $e^2 |\bar{r}_{1s,2p}|^2$ dependences.

C. (cont.)

(i) Evaluate $|r_{1s,2p}|^2 = e^2 |r_{2,1p}|^2 = e^2 |r_{2,1}|^2$,
 $e^2 |r_{2,1}|^2 = e^2 (2) \left(\frac{128}{243}\right) a_0^2$

Put $e = 1.602 \times 10^{-19} \text{ C}$

$a_0 = 5.292 \times 10^{-11} \text{ m}$

$e^2 |r_{2,1}|^2 \approx 3.988 \times 10^{-59} \text{ C}^2 \text{ m}^2$

(ii) Evaluate ω and the electric dipole moment
 $e |r_{2,1}| \approx 6.32 \times 10^{-30} \text{ Cm}$.

ω is defined as the angular frequency of the incoming photon, i.e.,

$\omega = \omega_{21} = \frac{E_2 - E_1}{\hbar}$, where E_1, E_2 are the energy levels of
1s and 2p states respectively.

Put $E_1 = -13.6 \text{ eV}$, $E_2 = -3.4 \text{ eV}$,

$\omega \approx 1.550 \times 10^{16} \text{ s}^{-1}$

Note: the photon is in ultraviolet region, slightly beyond the visible light region.

d. To compute the half life $\tau = \frac{1}{A}$ of a 2p state atom,

we need to compute A:

$A = \frac{\omega^3}{3\pi\epsilon_0\hbar^3} e^2 |r_{2,1}|^2$

Put the values computed in c):

$\omega \approx 1.550 \times 10^{16} \text{ s}^{-1}$

$e^2 |r_{2,1}|^2 \approx 3.988 \times 10^{-59} \text{ C}^2 \text{ m}^2$

↓ (cont.) and also put the value of constants:

$$h = 6.626 \times 10^{-34} \text{ J s}$$

$$c = 2.998 \times 10^8 \text{ m s}^{-1}$$

$$\epsilon_0 = 8.854 \times 10^{-12} \text{ m}^{-3} \text{ kg}^{-1} \text{ s}^4 \text{ A}^2$$

A is computed to be:

$$A = 3.132 \times 10^8 \text{ s}^{-1}$$

∴ The lifetime τ is:

$$\tau = \frac{1}{A} = 3.193 \times 10^{-9} \text{ s} //$$

Notes:

1. This is a typical lifetime for states with allowed electric dipole transitions downward.
2. We computed the value of A just based on the relation to $|t_{21}|$, i.e., electric dipole mechanism. In fact, this result turns out to be of the same order as the experiments.

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If we only have stimulated absorption and stimulated emission processes between two states 1 and 2, the dynamical equations

are :

$$\begin{cases} \frac{dN_1}{dt} = N_1 \cdot B_{12} U(\omega, T) \\ \frac{dN_2}{dt} = - N_2 B_{21} U(\omega, T) , \end{cases}$$

At equilibrium, the $1 \rightarrow 2$ transitions and $2 \rightarrow 1$ transitions are

balanced :

$$\frac{dN_1}{dt} = - \frac{dN_2}{dt}$$

$$\Rightarrow N_1 \cdot B_{12} U(\omega, T) = N_2 \cdot B_{21} U(\omega, T)$$

$$\Rightarrow N_1 = N_2 ,$$

where the stimulated absorption and emission rates are equal, i.e. $B_{12} = B_{21}$.

Hence in equilibrium, the number of atoms in two states

1 and 2 are equal, when the spontaneous emission is considered,

then $N_1 > N_2$

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Suppose we achieve $N_2 = 2N_1$ (in non-equilibrium situation) and

we use Eq. (2), which is valid in equilibrium physics, to evaluate the "temperature" of the system.

$$\frac{N_2}{N_1} = \frac{2N_1}{N_1} = e^{-\frac{(E_2 - E_1)}{kT}} = e^{-\frac{\hbar\omega}{kT}}$$

$$2 = e^{-\frac{2\pi\hbar c}{kT} (488 \text{ nm})}$$

$$T \approx -42535.2 \text{ K}$$

Negative temperature
Unreasonable results

Therefore, we can't have population inversion for equilibrium case.